

Alkali Emissions Measurements in Continuous CLC Operation



Ivan Gogolev

gogolev@chalmers.se

Project: Biomass Combustion Chemistry with Oxygen Carriers

Project Leader: Prof. Anders Lyngfelt

Funding:  Vetenskapsrådet



CHALMERS
UNIVERSITY OF TECHNOLOGY

Alkali Related Issues

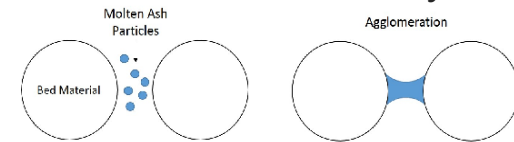
Gas-Phase Alkali Release

Fouling and corrosion of heat exchange equipment

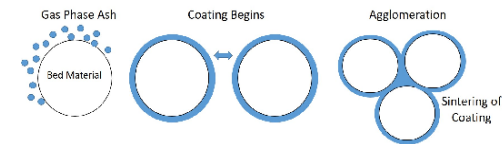


Alkalis in the bed material

Alkali silicates form sticky melts

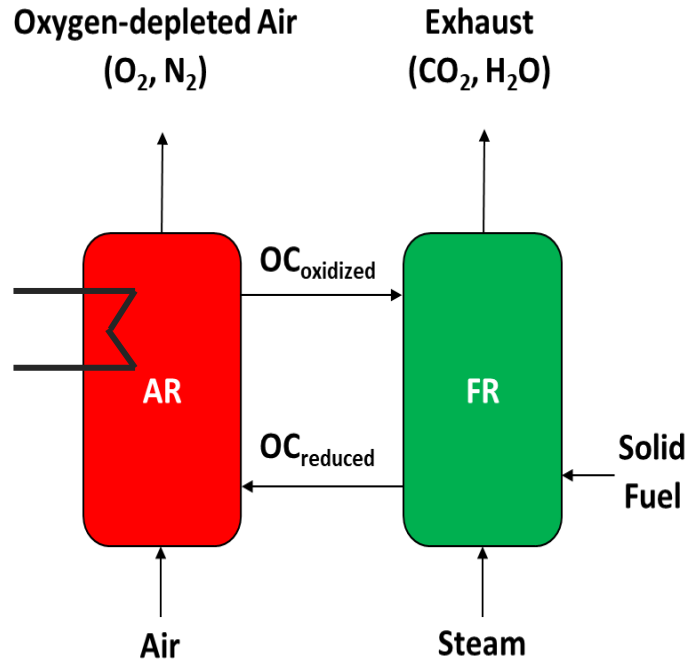


(a) Agglomeration from ash melting.



(b) Agglomeration from ash coating.

*Prominent modes of agglomeration
(Zimmerman et al. 2016)*

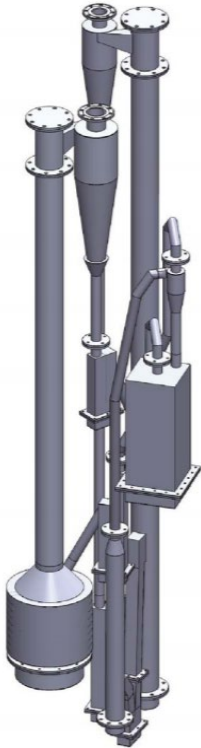


Main Hypotheses:

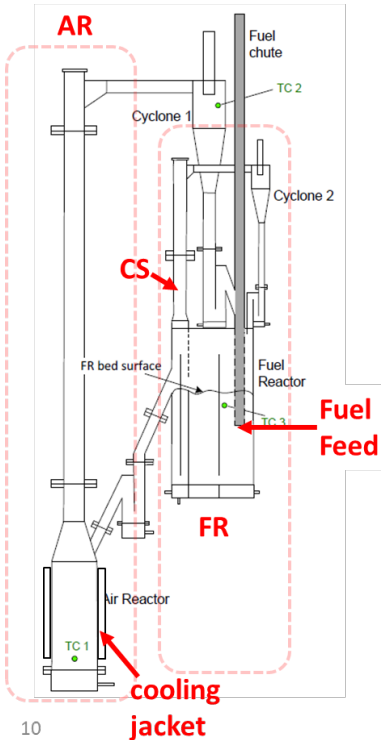
1. Most alkalis will be released to the gas phase in the FR
 - AR alkali release should be minimum
 - Less issues for HX surfaces (most process heat is recovered in the AR)
 - FR flue gas alkali will be easier to handle: lower flow rate, can operate FR HX at lower temp.
2. Overall gas-phase alkali release should be comparable to combustion of biomass

????

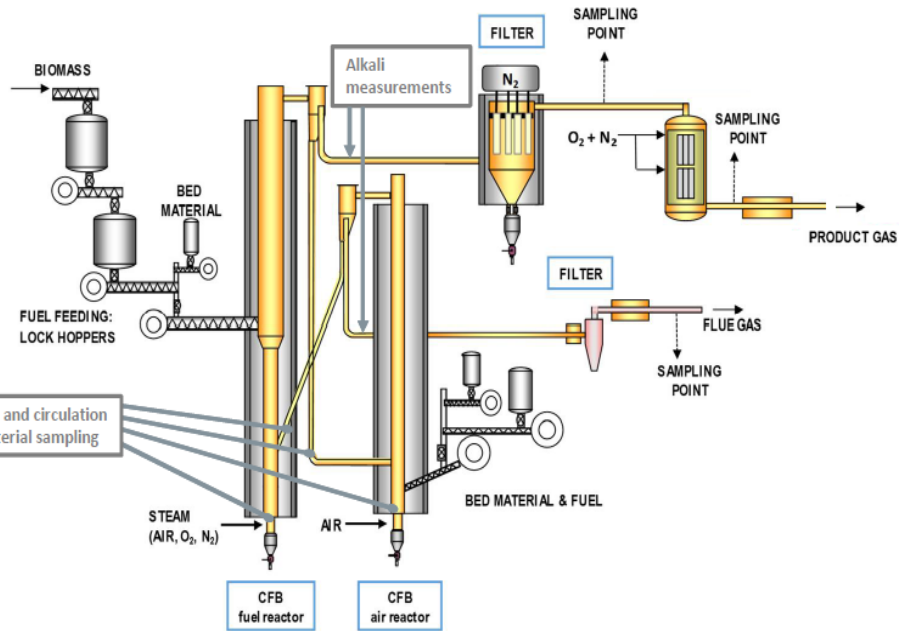
**Chalmers 100 kW CLC
Ilmenite & CaMn**



**Chalmers 10 kW CLC
Ilmenite**

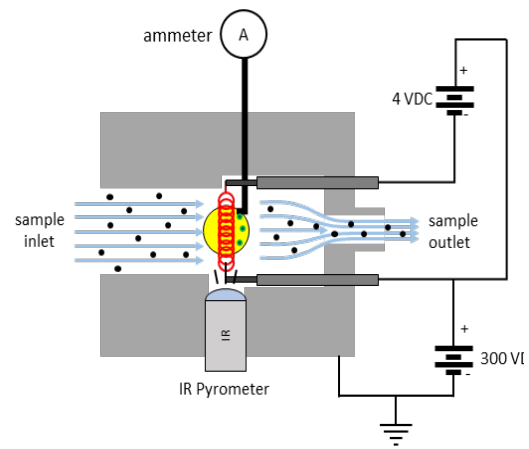


**VTT 60 kW D-CFB CLC
Ilmenite, Braunite**

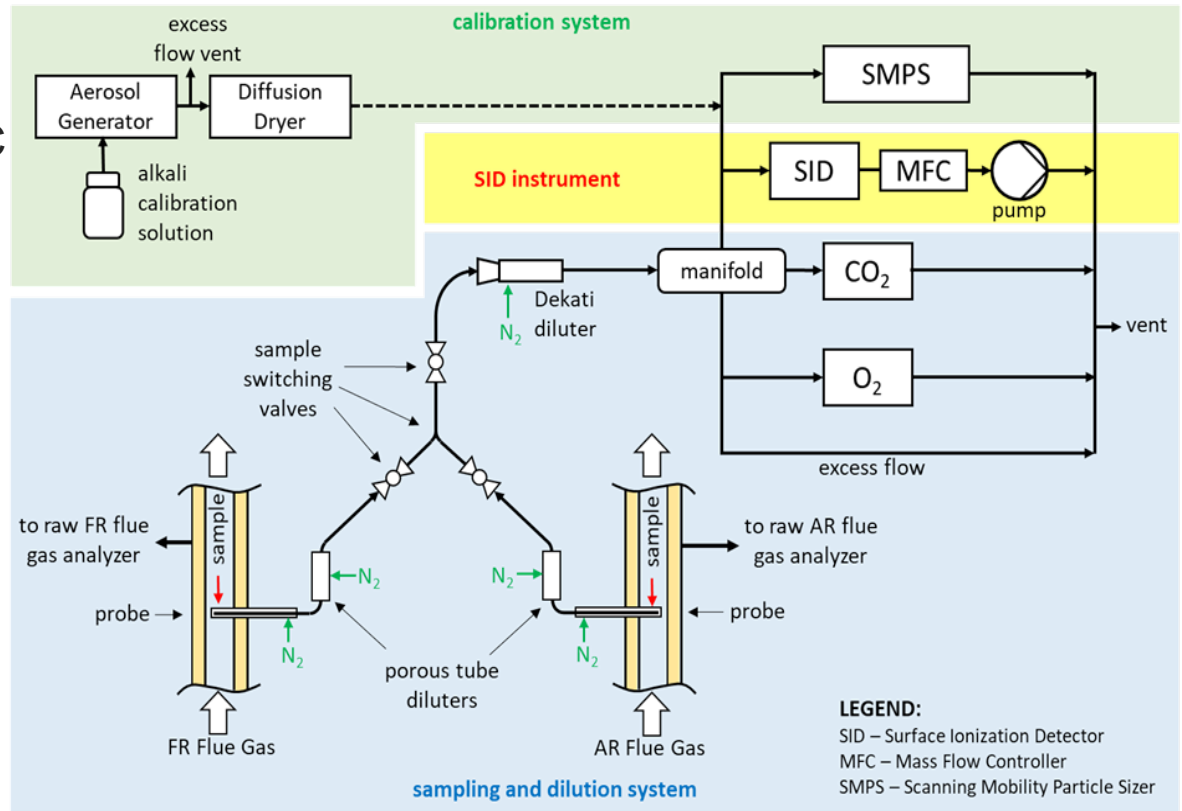


SID-based Alkali Measurement System

- Flue gas sampled at 900-1000° C
- 3 stages of sample dilution
- dilution tracking with CO₂/O₂



Surface Ionization Detector (SID)

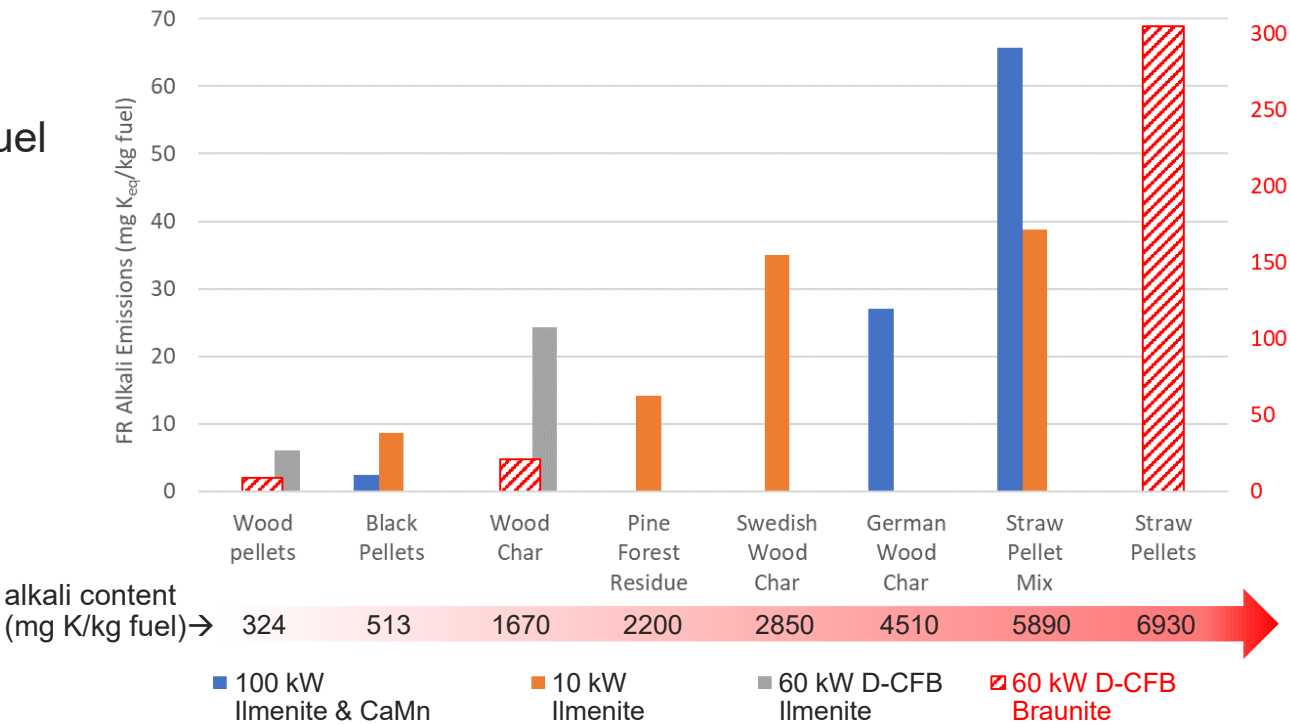


- Modular and transportable
 - Sampling from AR and FR
 - Dilution tracking for FR and AR
 - Optimized for CLC conditions
-
- First measurement of gas phase alkalis in CLC
 - Now tested at 5 different units



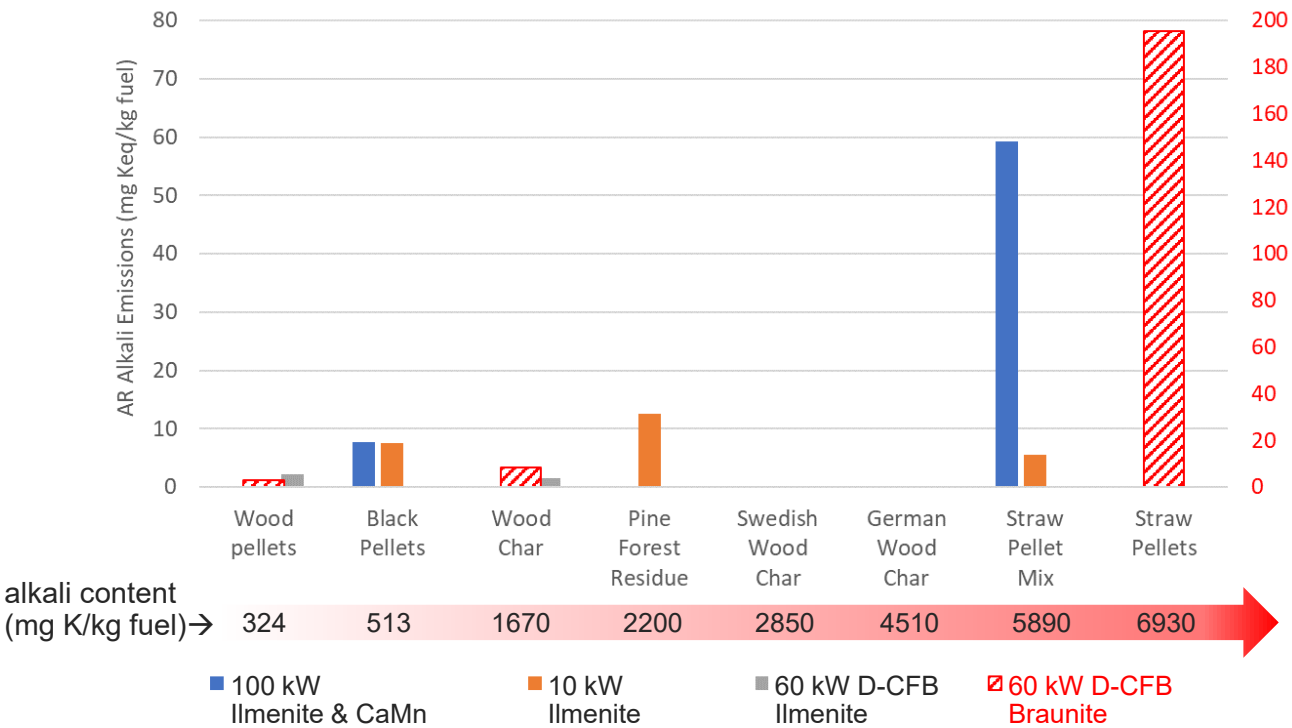
Fuel Reactor Emissions

- Emission increase with fuel alkali content
- Trend is consistent in all campaigns



Air Reactor Emissions

- No consistent trend



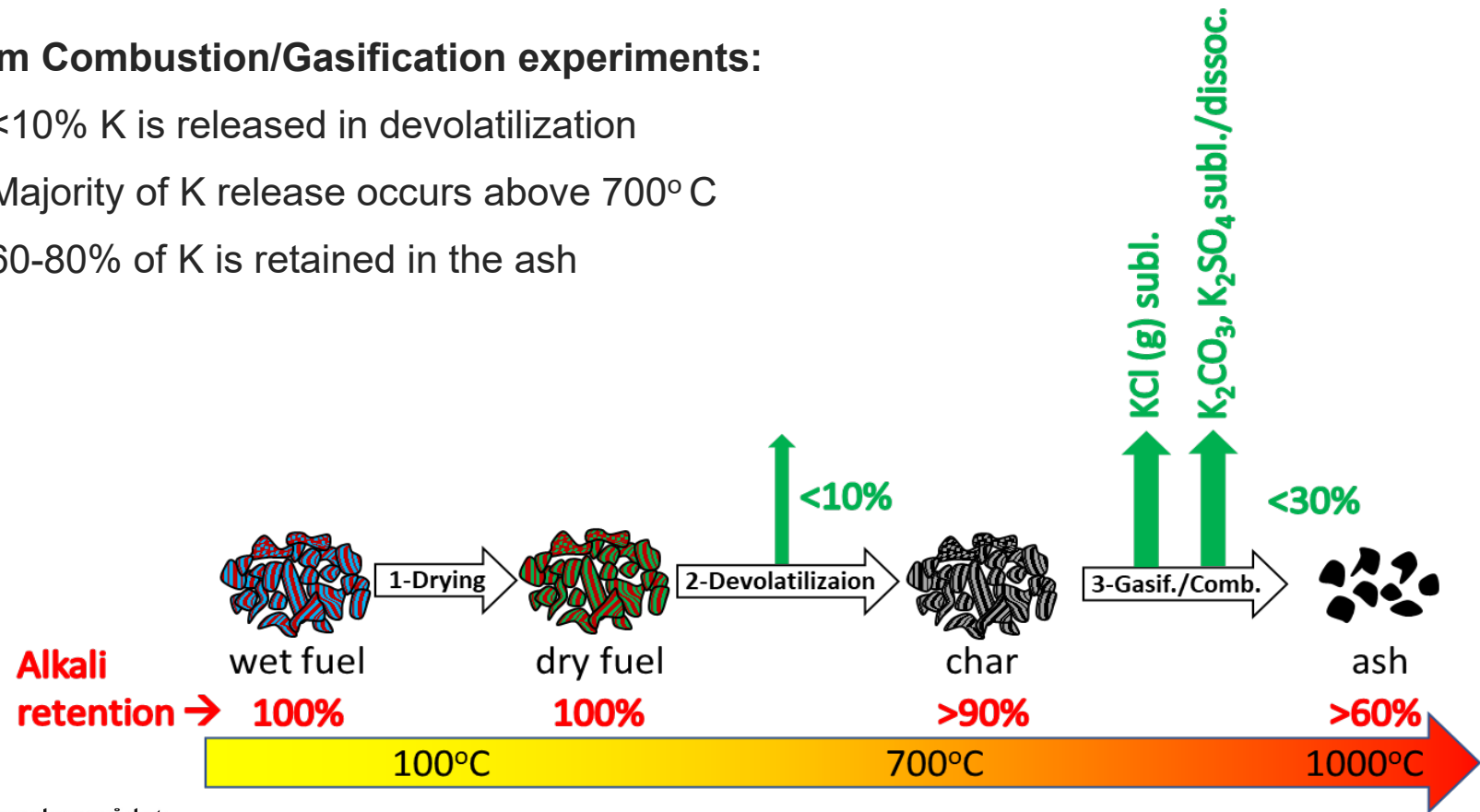
| Fuel | mg alkali/kg fuel | 100 kW Ilmenite & CaMn | | 10 kW Ilmenite | | 60 kW D-CFB Ilmenite | | 60 kW D-CFB Braunite | |
|---------------------|-------------------|---------------------------|------|-------------------|------|-------------------------|------|-------------------------|------|
| | | FR | AR | FR | AR | FR | AR | FR | AR |
| Wood Pellets | 324 | | | | | 1.9% | 0.7% | 2.9% | 1.0% |
| Black Pellets | 513 | 0.7% | 1.8% | 1.7% | 1.5% | | | | |
| Wood Char | 1670 | | | | | 1.3% | 0.1% | 1.3% | 0.5% |
| Pine Forest Residue | 2200 | | | 0.7% | 0.6% | | | | |
| Swedish Wood Char | 2850 | | | 1.2% | | | | | |
| German Wood Char | 4510 | | | | | | | | |
| Straw Pellet Mix | 5890 | 1.1% | 1.1% | 0.7% | 0.1% | | | | |
| Straw Pellets | 6930 | | | | | | | 4.4% | 2.8% |

Alkali Emissions Distribution:

- 0.8 – 7.2 wt% of fuel alkalis are released to the gas phase
- AR emissions can be significant
- Several cases where AR emissions = or > FR emissions

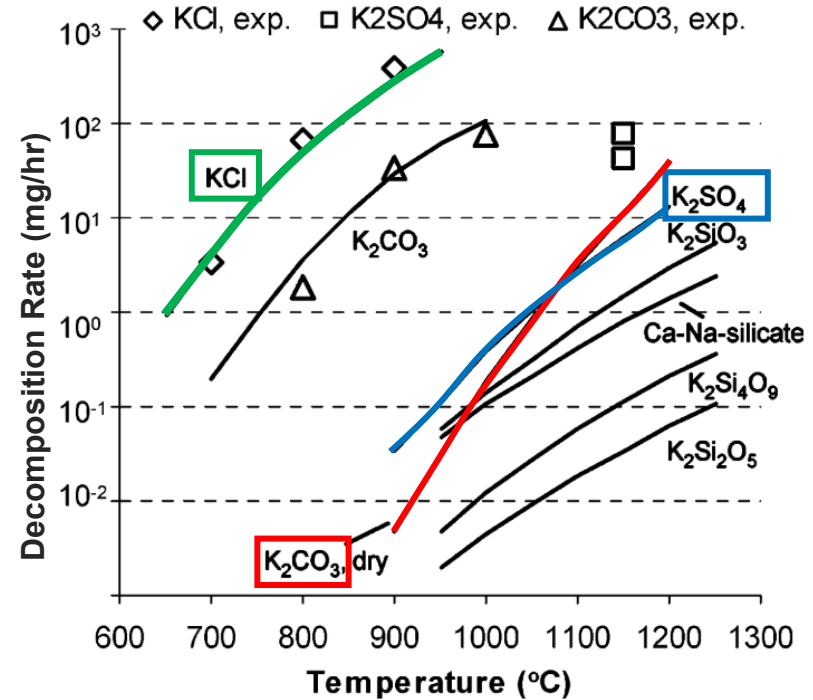
From Combustion/Gasification experiments:

- <10% K is released in devolatilization
- Majority of K release occurs above 700° C
- 60-80% of K is retained in the ash



Carryover of condensed alkalis to AR can occur via carryover of:

1. Unconverted char
 2. Ash
- Both can contain undecomposed K compounds
 - Continue gas phase release of alkali in the AR
 - AR temperature is up to 100° C higher than the FR



Potassium compound release rates vs. Temperature (Knudsen et al. 2004)

| Fuel | mg alkali/kg fuel | 100 kW Ilmenite & CaMn | 10 kW Ilmenite | 60 kW D-CFB Ilmenite | 60 kW D-CFB Braunite |
|---------------------|-------------------|---------------------------|-------------------|-------------------------|-------------------------|
| Wood Pellets | 324 | | | 97.5% | 96.9% |
| Black Pellets | 513 | 97.5% | 96.8% | | |
| Wood Char | 2200 | | 98.7% | | |
| Pine Forest Residue | 1670 | | | 98.6% | 98.2% |
| Swedish Wood Char | 2850 | | | | |
| German Wood Char | 4510 | | | | |
| Straw Pellet Mix | 5890 | 97.8% | 99.2% | | |
| Straw Pellets | 6930 | | | | 92.8% |

Alkali Retention:

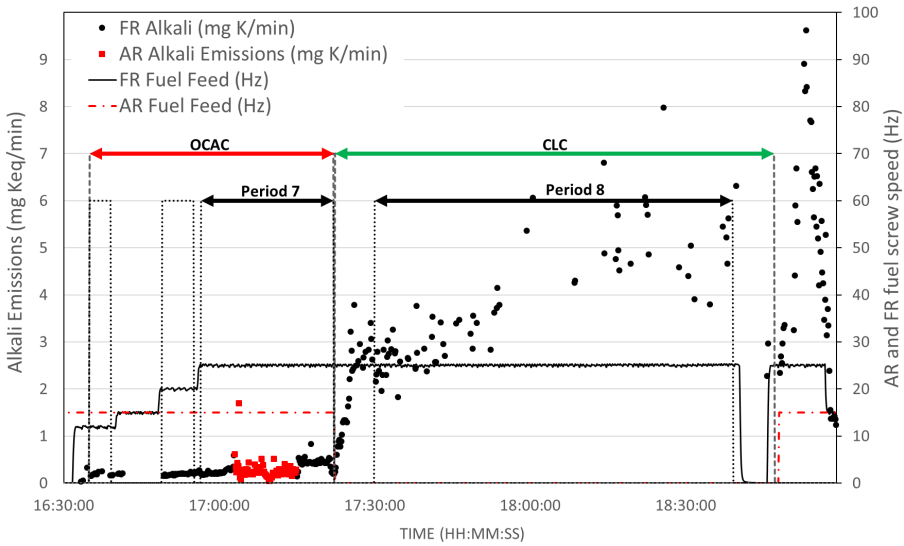
- Retention is >92% for all oxygen carriers, fuels, and tests
- Retention was confirmed by bed material and filter material analysis

Retention can be divided into

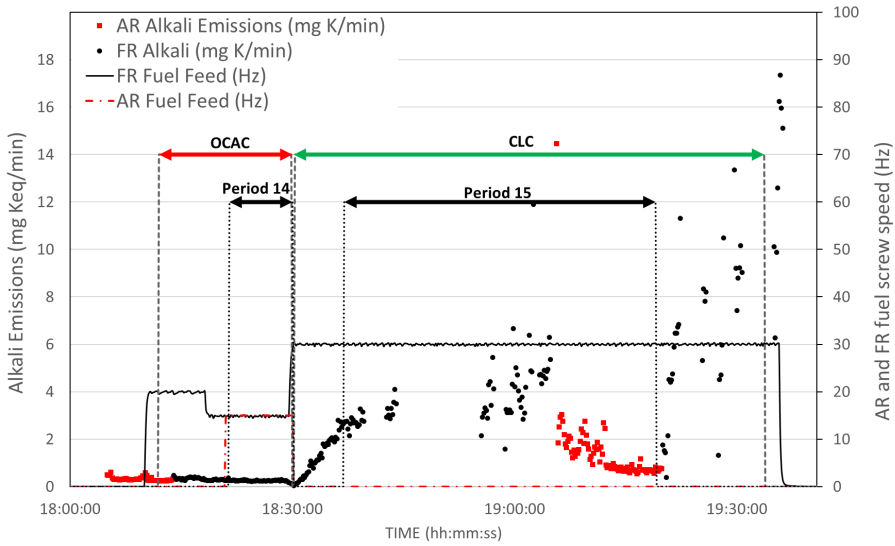
1. Retention by fuel ash formation process → estimated at 60-80 wt%
2. Effect of OC material → 10-30 wt%

Both Ilmenite and Braunite can form the following stable compounds at CLC temperatures:

- K-silicates
- K-aluminosilicates
- K-manganates
- Ilmenite can also form K-titanates



Alkali emissions (Ilmenite & Wood Char)



Alkali emissions (Braunitite & Wood Char)

| OC Material | Fuel | mg alkali/kg fuel | OCAC FR emissions (mg K/kg fuel) | CLC FR emissions (mg K/kg fuel) | % increase with CLC |
|-------------|---------------|-------------------|-------------------------------------|------------------------------------|---------------------|
| Ilmenite | Wood Pellets | 324 | 1.9 | 6.1 | 217% |
| | Wood Char | 1670 | 2.3 | 24.4 | 983% |
| Braunite | Wood Pellets | 324 | 9.2 | 8.7 | -5% |
| | Wood Char | 1670 | 3.2 | 21.4 | 564% |
| | Straw Pellets | 6930 | 19.4 | 305.3 | 1470% |

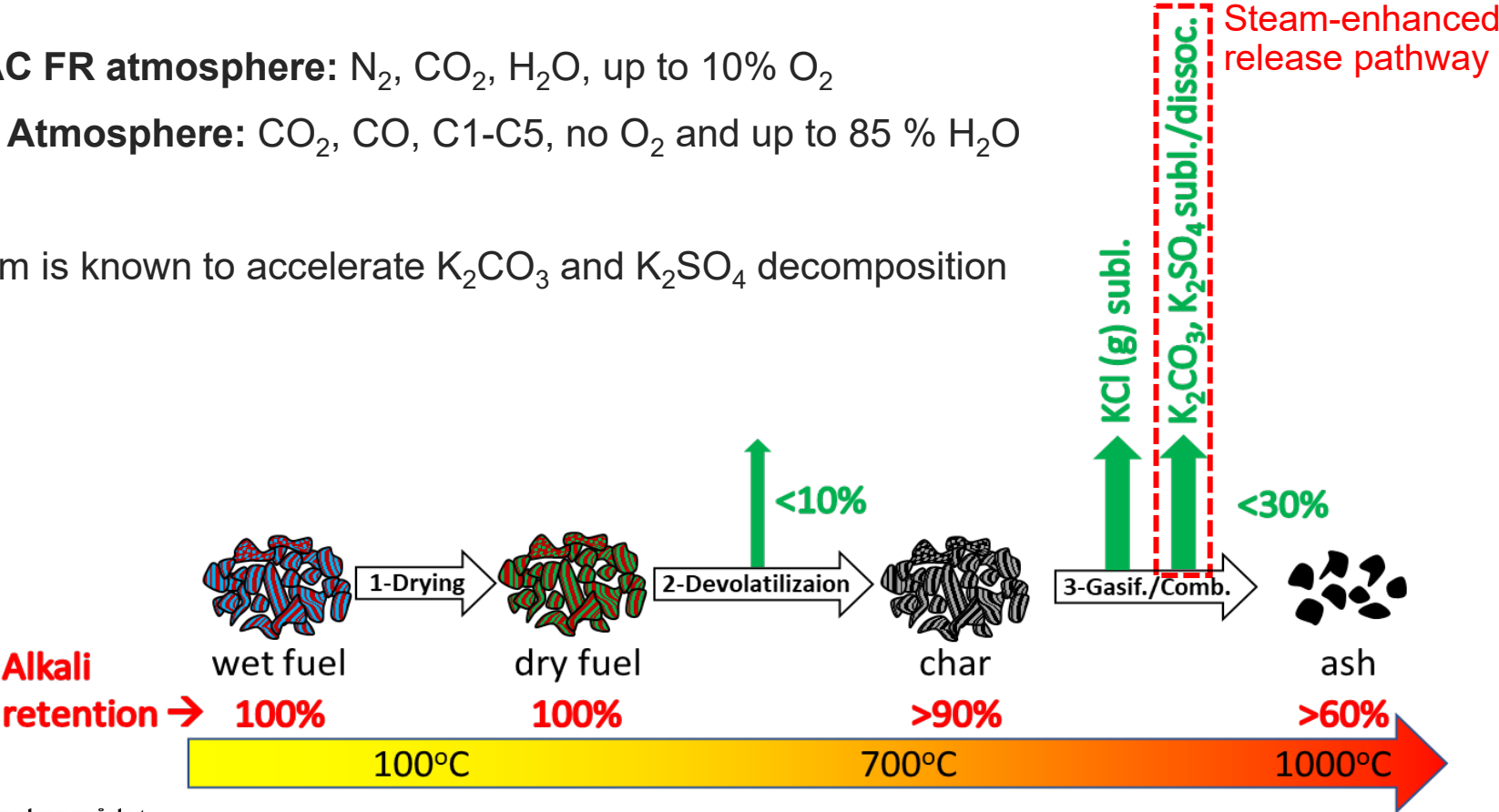
Effect of FR atmosphere:

- FR emissions in CLC mode higher than in OCAC mode
- Difference cannot be explained by temperature difference of the two modes

OCAC FR atmosphere: N₂, CO₂, H₂O, up to 10% O₂

CLC Atmosphere: CO₂, CO, C1-C5, no O₂ and up to 85 % H₂O

Steam is known to accelerate K₂CO₃ and K₂SO₄ decomposition



CONCLUSIONS

- SID based measurements are well suited for alkali measurement in CLC
- <10% of alkalis are release to the gas phase in CLC
- FR alkali emissions are typically higher than AR, but AR emissions can be significant
- >90% of alkalis are retained in condensed form
- CLC operation releases more gas-phase alkalis vs. OCAC operation
- FR's steam content is responsible for increased emissions levels

Thank You !