

Chemical-Looping Combustion of Solid Fuels

-

Status and recent progress



Anders Lyngfelt



GHGT-13

Lausanne, November 15-18, 2016

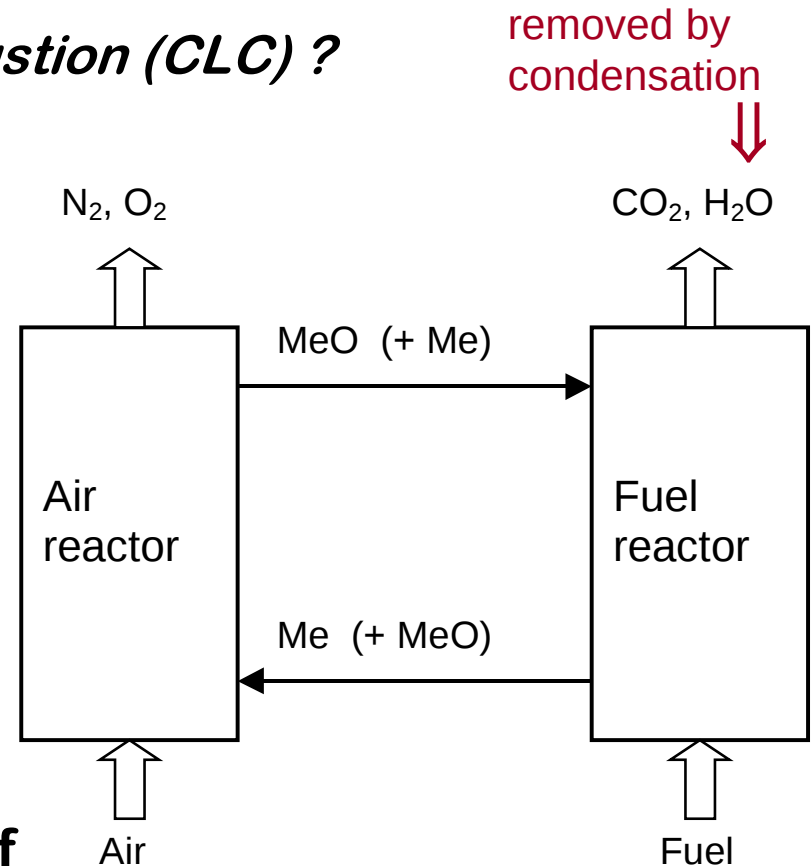
Why chemical-looping combustion (CLC) ?

Oxygen is transferred from air to fuel by metal oxide particles

Inherent CO₂ capture:

- fuel and combustion air *never mixed*
- *no active gas separation needed*
- large costs/energy penalties of gas separation avoided

- **Potential for real breakthrough in costs of CO₂ capture**



The oxygen carrier is the cornerstone of CLC



Overview CLC operation (150 publications)

	Reported operational time, h	Of which solid fuels
<u>Manufactured materials:</u>		
Nickel	3167	267
Copper	1036	158
Manganese	91	0
Iron	1652	1070
Cobalt	178	0
Combined oxides	718	106
<u>Ores or waste materials:</u>		
Iron	847	624
Ilmenite	1163	717
Manganese	243	158
Calcium sulfate	75	75
Total Manufactured	6842	1601
Total ores/waste	2238	1574
Total	9170	3175

>70 materials

Oxygen carriers

- Long experience of operation of a number of materials, in a number of different pilots, provides proof-of-concept
- Low cost materials (ores of ilmenite, manganese or iron) can be used with solid fuels, but less suitable for methane-rich gaseous fuels

Overview CLC units

Location	Unit	Oxides tested	Fuel\references	Year
Chalmers	10 kW-GL	NiO, Fe ₂ O ₃ , CaMnO ₃ , ilmenite	nat. gas, oil ^{1 2}	2004
KIER	50 kW	NiO, CoO	nat. gas ³	2004
CSIC	10 kW	CuO, NiO	nat. gas ⁴	2006
Chalmers	0.3 kW	NiO, Mn ₃ O ₄ , Fe ₂ O ₃ , CuO, ilmenite, CaMnO ₃ , Mn/Fe, Mn/Mg, Mn/Si, Mn/Fe/Si, Mn ore	nat. gas, syngas ⁵	2006
Chalmers	10 kW-SF	ilmenite, Mn ore, CaMnO ₃	coal, petcoke ^{6 7}	2008
CSIC	0.5 kW-GL	CuO, NiO, Fe ₂ O ₃ , CaMnO ₃ ,	nat. gas, acid gas, sour gas, ethanol ⁸	2009
KAIST	1 kW	NiO + Fe ₂ O ₃	CH ₄ ⁹	2009
Vienna UT	140 kW	ilmenite, NiO, CuO	nat. gas, CO, H ₂ ¹⁰	2009
Alstom	15 kW	NiO	nat. gas ¹¹	2009
Nanjing	10 kW-SF	NiO, Fe ₂ O ₃	coal, biom. ¹²	2009
KIER	50 kW	NiO, CoO	nat.gas, syngas ¹³	2010
Nanjing	1 kW-SF	Fe ₂ O ₃ (ore)	coal, biomass, sewage sludge ^{14 15}	2010
IFP-Lyon	10 kW-GSF	NiO, CuO, Mn ore	CH ₄ , coal, syngas ^{16 17}	2010
Stuttgart	10 kW	ilmenite	syngas ¹⁸	2010
Xi'an Jiaotong	10 kW- Pr	CuO/Fe ₂ O ₃	coke oven gas ¹⁹	2010
CSIC	1.5 kW-SF	ilmenite, CuO, Fe ₂ O ₃	coal ²⁰	2011
Chalmers	0.3 kW LF	NiO, Mn ₃ O ₄ , CuO	kerosene ²¹	2011
Chalmers	100 kW-SF	ilmenite, Fe ore, Mn ore	coal, petcoke, wood char ^{22 23}	2012
Hamburg	25 kW-SF	ilmenite	coal, CH ₄ ²⁴	2012
Ohio	25 kW-SF	Fe ₂ O ₃	coal ^{25 26}	2012
Nanjing	50 kW-Pr-SF	iron ore	coal ²⁷	2012
Tsinghua	0.2 kW	ilmenite	CO ²⁸	2013
Darmstadt	1 MW-SF	ilmenite	coal ^{29 30}	2015
Alstom	3 MW-SF	CaSO ₄ /CaS	coal ³¹	2014
CSIC	50 kW-SF	ilmenite, iron ore, CuO/Fe ₂ O ₃ /MgAl ₂ O ₄	coal, lignite, anthracite ³²	2014
Huazhong	5 kW-GSF	iron ore	CH ₄ coal ³³	2015
Guangzhou	10 kW-G	Fe ₂ O ₃	saw dust ³⁴	2015
Nanjing	25 kW-G	NiO, iron ore	rice husk ³⁵	2015
KIER	200 kW	NiO	nat. gas ³⁶	2016
Huazhong	5 kW-SF	iron ore	coal ³⁷	2016
Sintef	150 kW	CuO	CH ₄ ³⁸	2016
VTT	20 kW-SF	ilmenite	biomass ³⁹	2016
NETL	50 kW	CuO/Fe ₂ O ₃	CH ₄ ⁴⁰	2016
Chalmers	1.4/10 MW-SF	ilmenite, Mn ore	biomass ⁴¹	2016

SF-solid fuel, GSF-gaseous & solid fuel, Pr-pressurized, LF-liquid fuel,
GL=gaseous/liquid fuel, G-Gasification

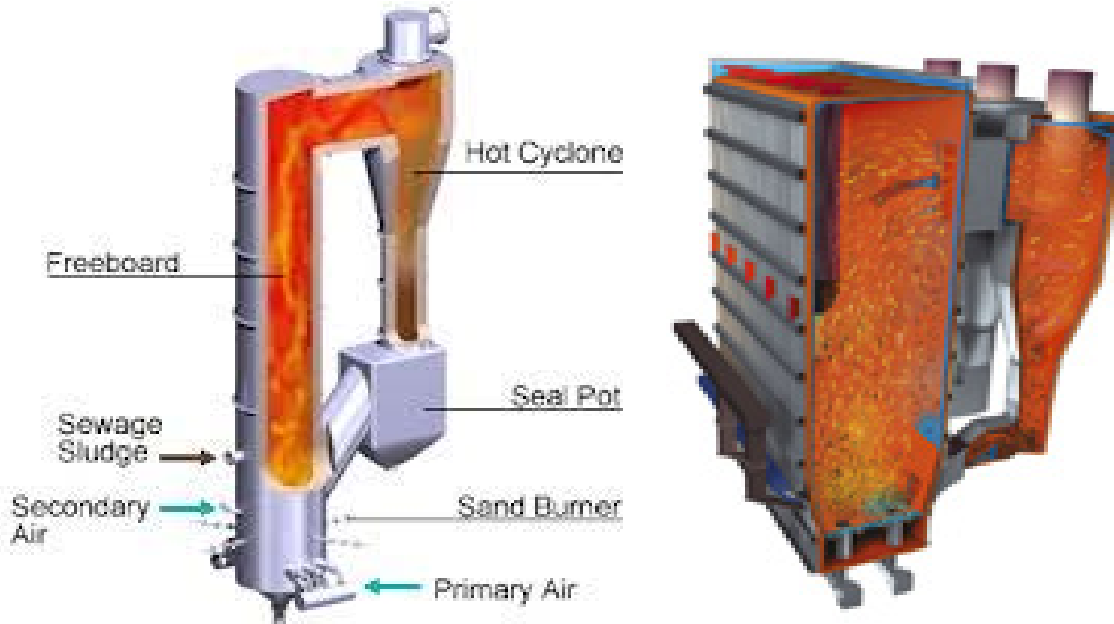
34 CLC units

18 units for solid fuels
(including 2 gasifiers)

Size range 0.2 kW – 3 MW

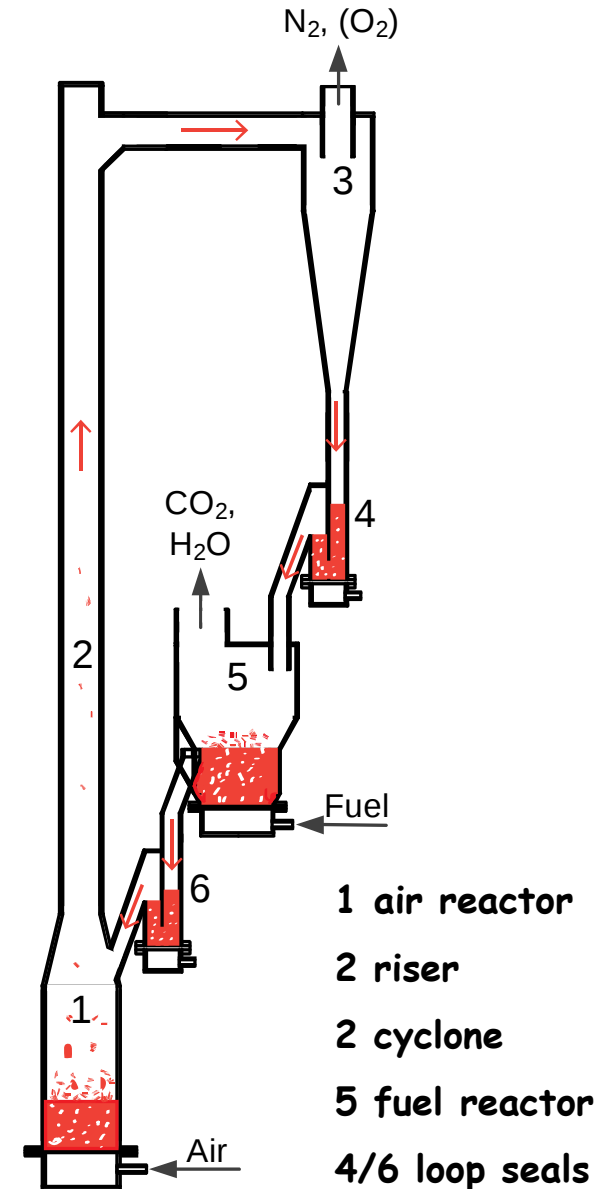
Dual fluidized beds, except
Ohio, moving bed.

Circulating fluidized bed (CFB) combustion technology is common for burning solid fuels.

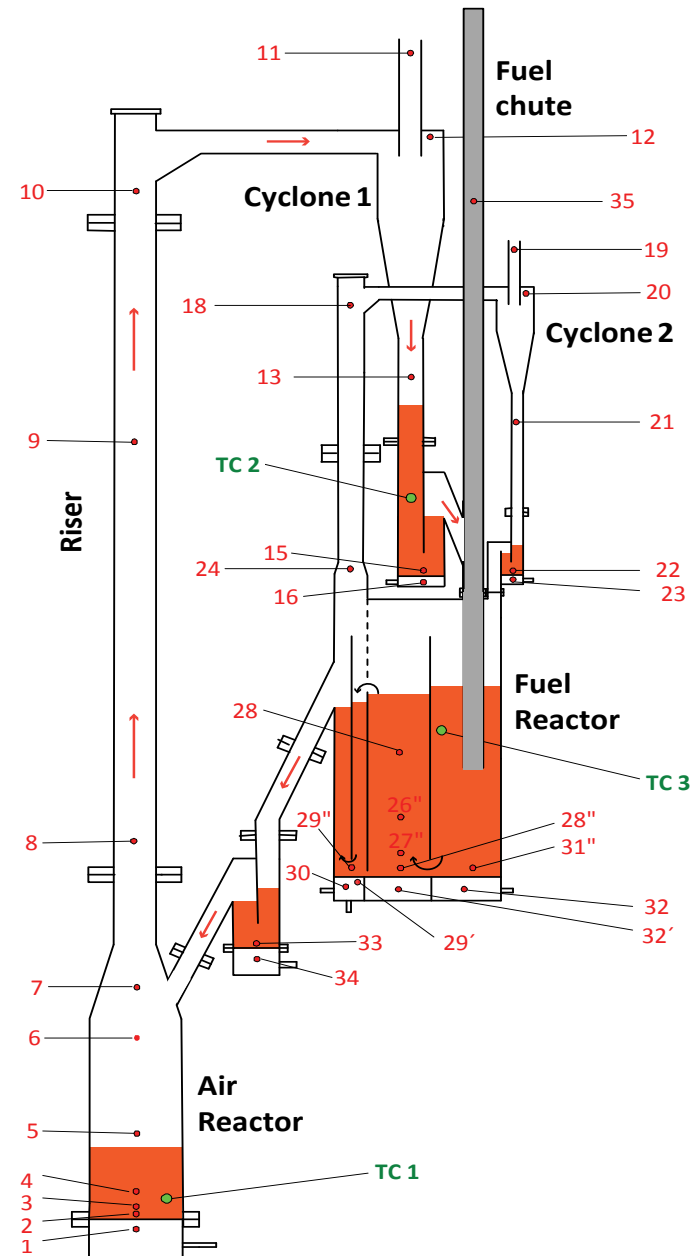


CLC uses CFB principle to circulate bed material. The principal difference is adding a fuel reactor on the return side.

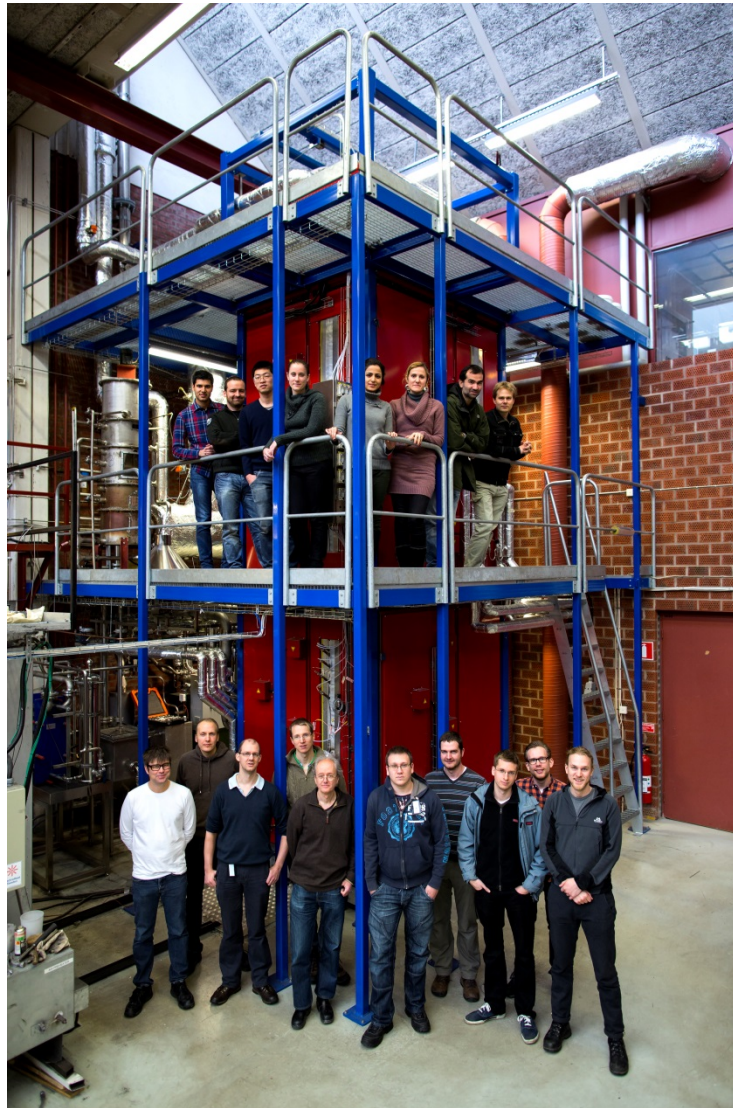
Example: 10 kW CLC pilot for gaseous fuel (2003)



Chalmers' 10 kW chemical-looping combustor for solid fuels, 2006



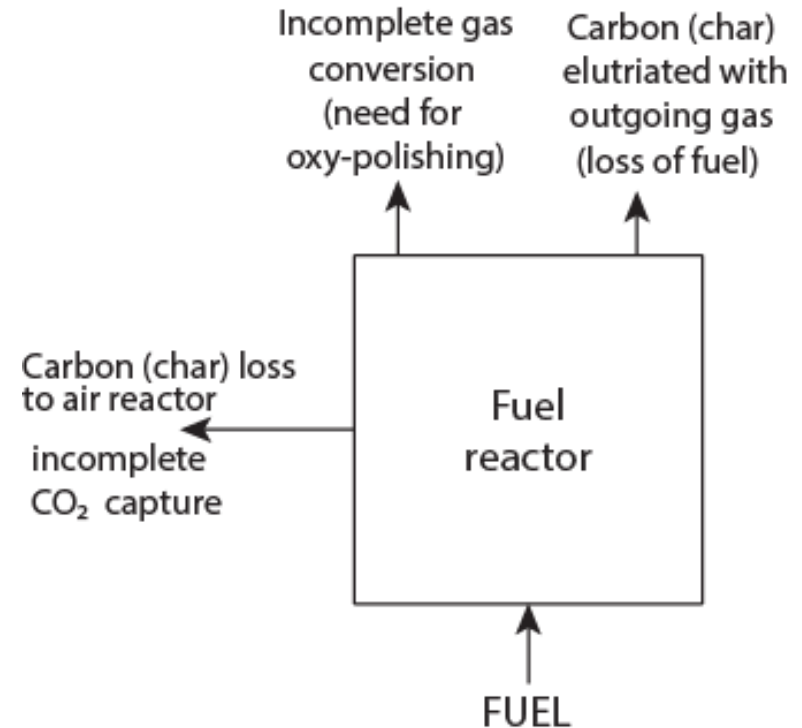
Chalmers' 100 kW CLC for solid fuel, publ. 2012



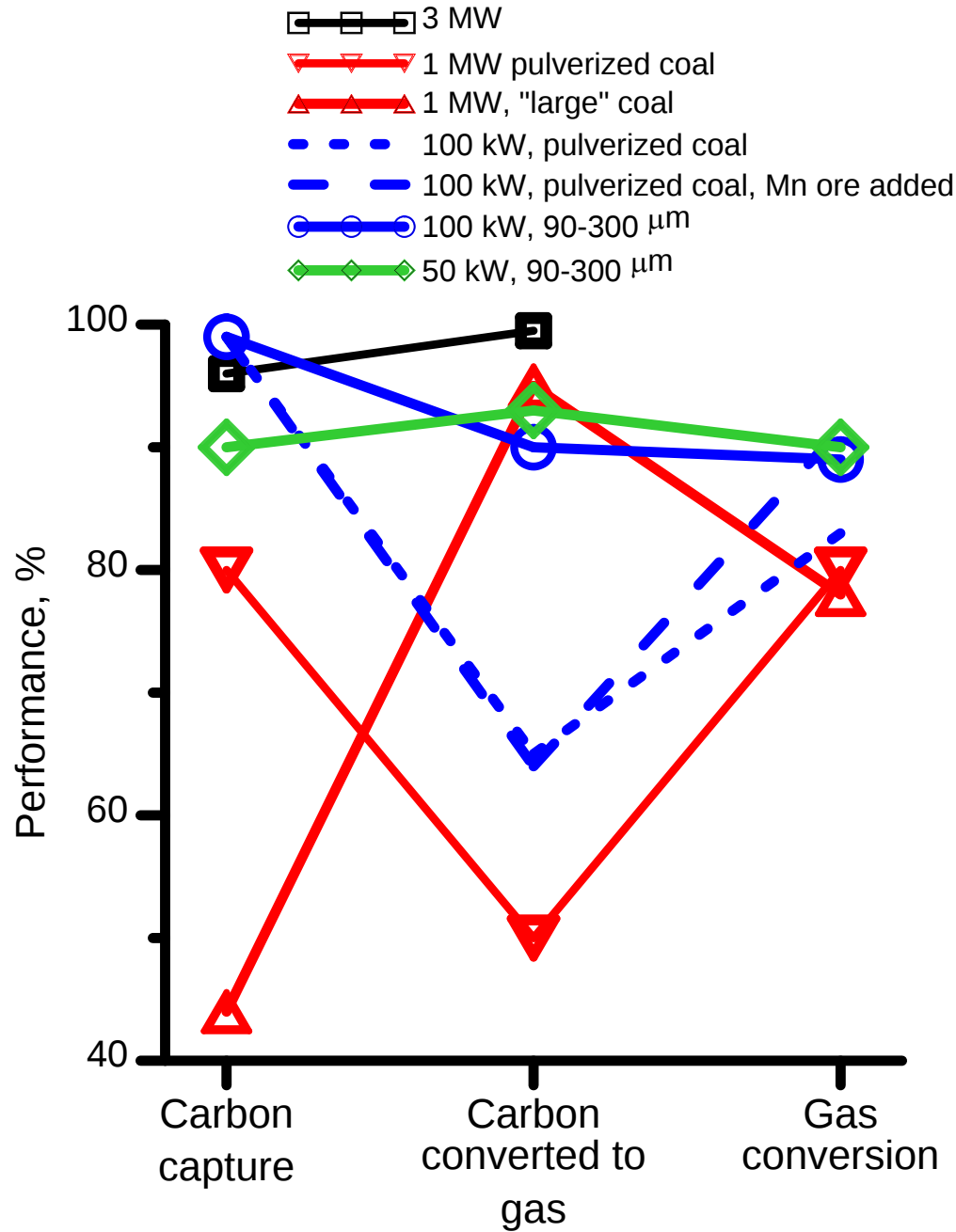
Operation with solid fuels –

3 causes for incomplete fuel conversion

- Gas not fully oxidized to $\text{CO}_2/\text{H}_2\text{O}$
 - oxygen needs to be added after fuel reactor
 - Fine char particles lost from fuel reactor
 - added cost of fuel, increased waste disposal.
 - Char follows particle flow to air reactor.
 - CO_2 capture not complete.
- ⇒ 3 performance criteria:
- ⇒ 1) gas conversion
 - ⇒ 2) carbon conversion to gas
 - ⇒ 3) CO_2 capture



Performance in pilot operation



Fuel size:

- Avoid fines, e.g. pulverized coal, for low carbon loss
- Avoid larger particles, e.g. pulverized coal, for high CO₂ capture
- Intermediate sized coal, 100-300 mm, likely optimal

Scale-up

Full scale expected to :

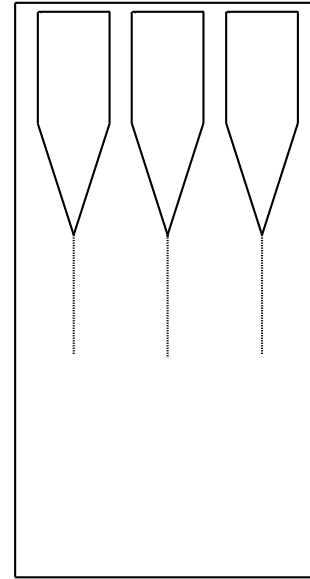
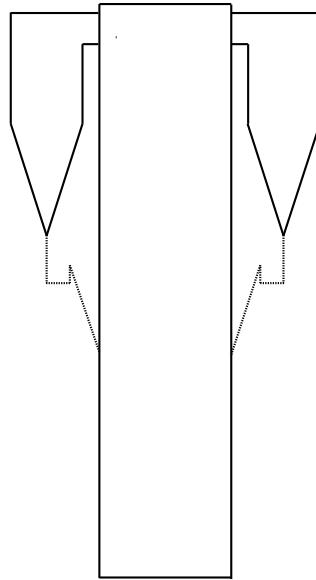
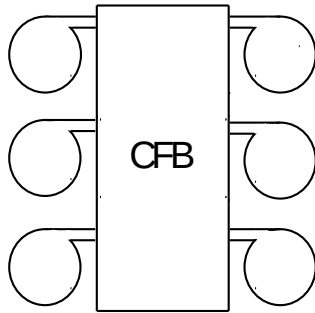
- reduce loss of carbon fines significantly because of increased residence time
 - High riser
 - Better cyclone efficiency
- improve gas-solids contact in riser
- reduce gas-solids contact in bottom bed

But ...

Incomplete gas conversion difficult to avoid

- Downstream oxygen polishing likely best way to reach full conversion
- Oxygen need expected to be 5-10 times less than in oxyfuel

1000 MW_{th}
CFB boiler
dimensions
11x25.5x48

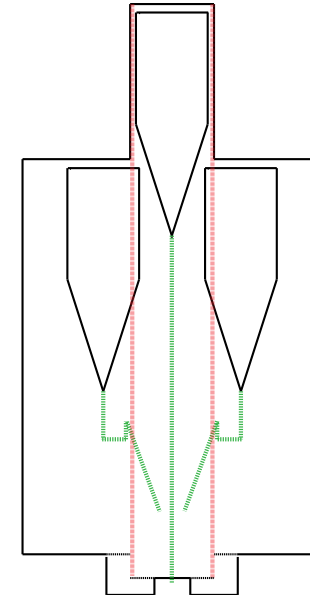
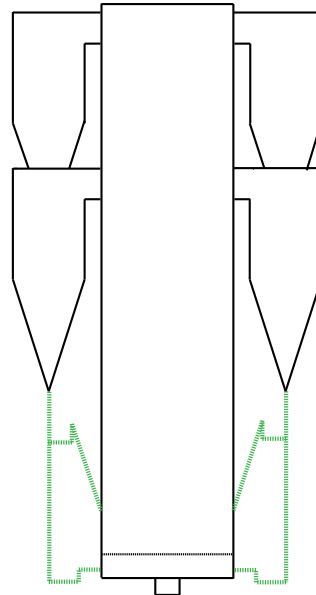
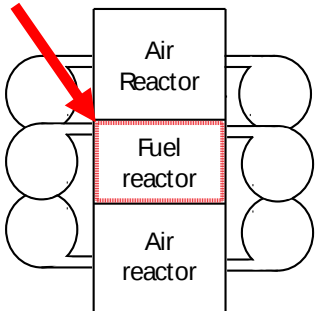


**Fuel reactor,
cyclones, ducts and
post-oxidation
chamber: 2500 m²**

Cost: 1500 €/m²

1000 MW_{th}
CLC boiler
dimensions
11x25x48

**Added cost:
insulation of
fuel reactor**



**Added cost of fuel
reactor:**

4 M€

⇒ 0.4 M€/year

÷

2 Mton CO₂/year

= 0.2 €/ton CO₂

Other costs

- **CO₂ compression**
 - Similar to other capture technologies
- **Oxygen production (incomplete conversion)**
 - 5-10 times less oxygen compared to oxyfuel
- **CO₂ purification**
 - As in oxyfuel, option for SO₂/NO_x capture
- **Oxygen carrier**
 - With low cost ores, estimated to 1-4 €/tonne CO₂
- **Minor costs, >1 €/tonne**
 - Fuel grinding, steam for fluidization
- **Total costs, estimated to 16-26 €/tonne CO₂**

Strategy for full-scale demonstration of chemical-looping at low cost ?

Build dual purpose CFB/CLC, or retrofit CFB to CLC

- **Low added cost of CLC plant**

Skip CO₂ capture (in 1st stage)

- **Major added costs can be avoided, i.e. CO₂ compression and purification, and oxygen production**

Go for biomass

- **Potential advantages for avoiding fouling/high-temperature corrosion, thus potential of higher steam data/higher efficiency. Pollutants (NO_x) in smaller CO₂ stream, emissions can be reduced**

When the technology successfully demonstrated, add CO₂ capture (2nd stage)

Conclusions

- >9000 h of operation and >3000 h with low-cost mtrls shows CLC is feasible
 - Good performance can be reached in pilots, and expected to improve in large scale
 - Incomplete gas conversion can be addressed by "oxy-polishing"
- **Significant cost reduction because little gas separation needed**
- *Technology is ready for scale-up !*

THANK YOU !
QUESTIONS ?

>300 CLC publications at:

<http://www.entek.chalmers.se/lyngfelt/co2/co2publ.htm>

