

Role of ash chemistry in

CHEMICAL LOOPING WITH BIOMASS

virtual seminar to bring together big ideas

MAY | 25-26 | 2021

9:00AM-12:30PM (CET, GMT+2)



CHALMERS
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Vetenskapsrådet
Swedish Research Council

CSIC expertise on Biomass Chemical Looping Gasification (BCLG)

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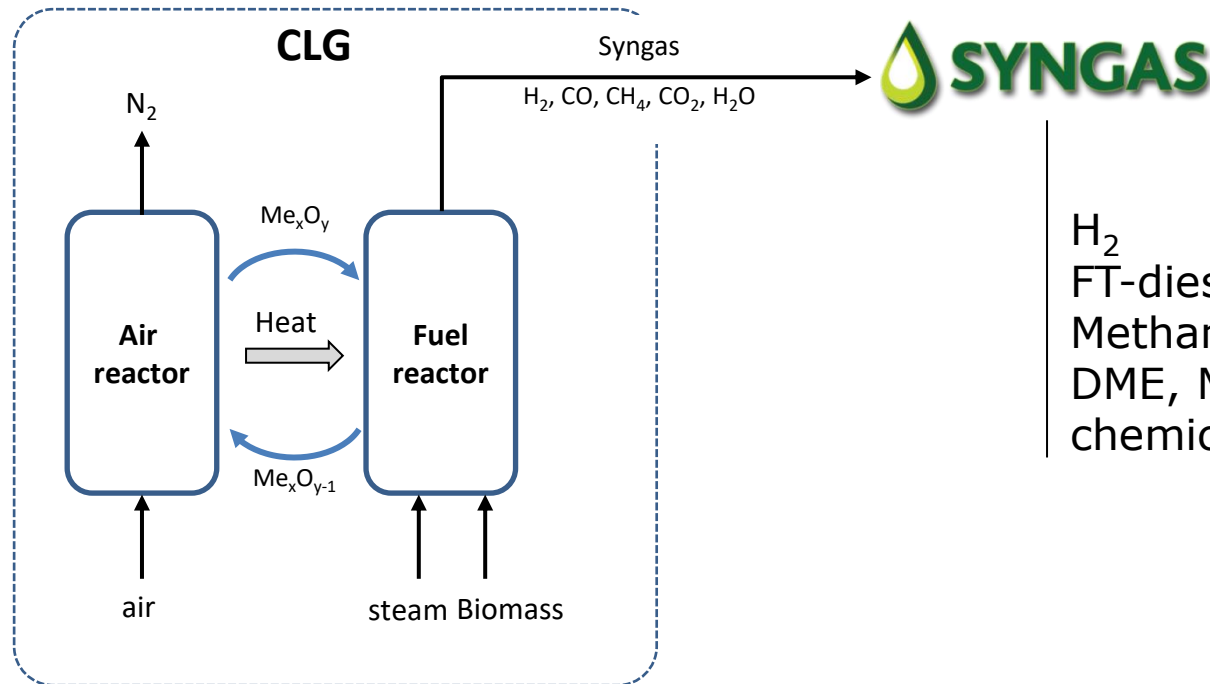
Outline

- General CLG concept
- Control of oxygen in BCLG.
- Summary of BCLG tests at CSIC
- Experience in the 50 kW_{th} CLG unit regarding agglomeration

Chemical Looping Gasification



Biomass, wastes



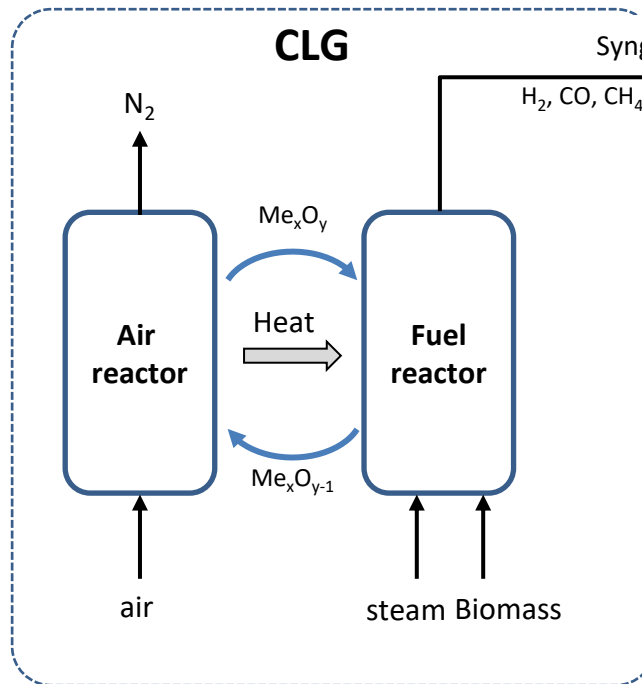
H₂
FT-diesel, gasoline
Methanol
DME, MTBE
chemicals

- Pure syngas stream (no N₂ dilution and without using ASU unit)
- Autothermic
- Less steam consumption than conventional gasification processes
- Low tar generation (catalytic properties of the Me_xO_y)
- Less emission of CO₂ to the atmosphere (CO₂ concentrated in syngas)
- Less corrosion in heat exchangers (located in AR)

Chemical Looping Gasification



Biomass, wastes



 SYNGAS

H₂
FT-diesel, gasoline
Methanol
DME, MTBE
chemicals

Oxygen-to-biomass ratio, λ

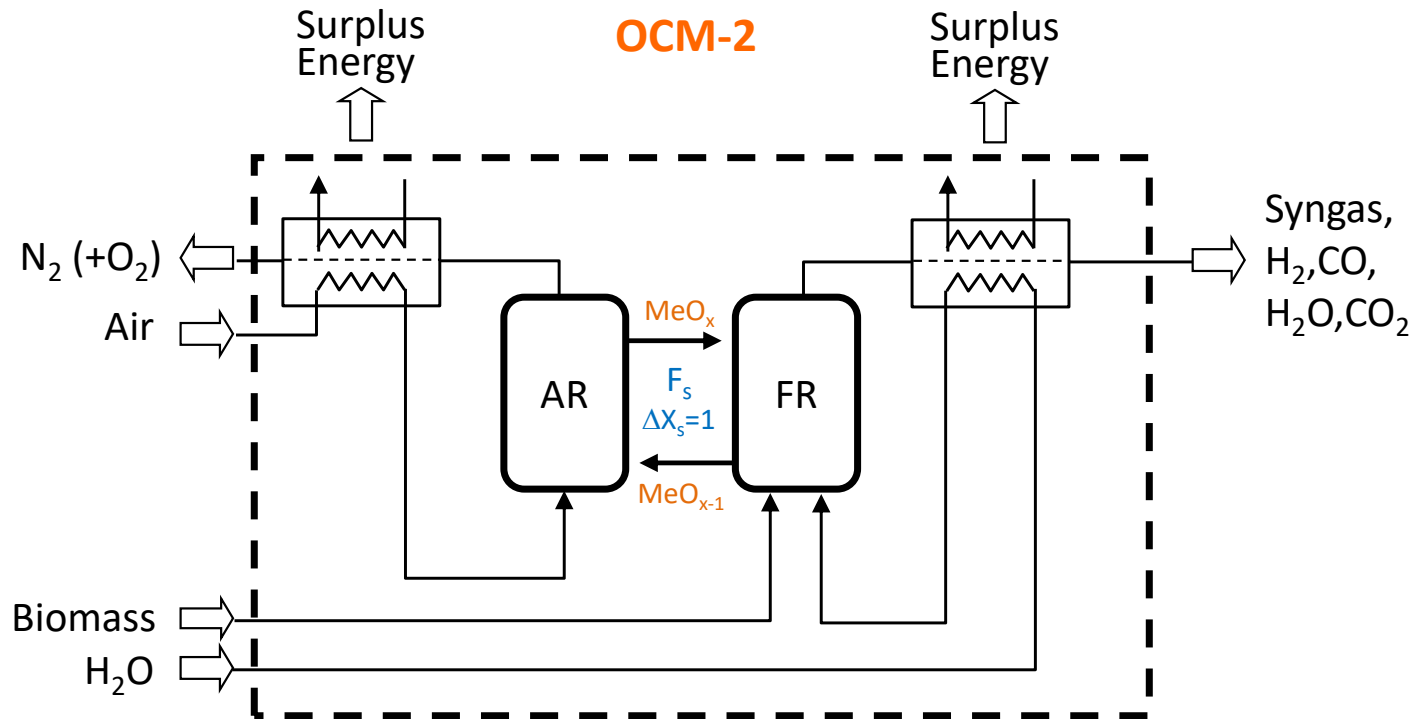
$$\lambda = \frac{\text{oxygen fed in the AR (mol/h)}}{\text{oxygen needed for full biomass combustion (mol/h)}}$$

Steam to Biomass ratio, S/B

$$S/B = \frac{\text{flow of steam (kg/h)}}{\text{flow of dry biomass (kg/h)}}$$

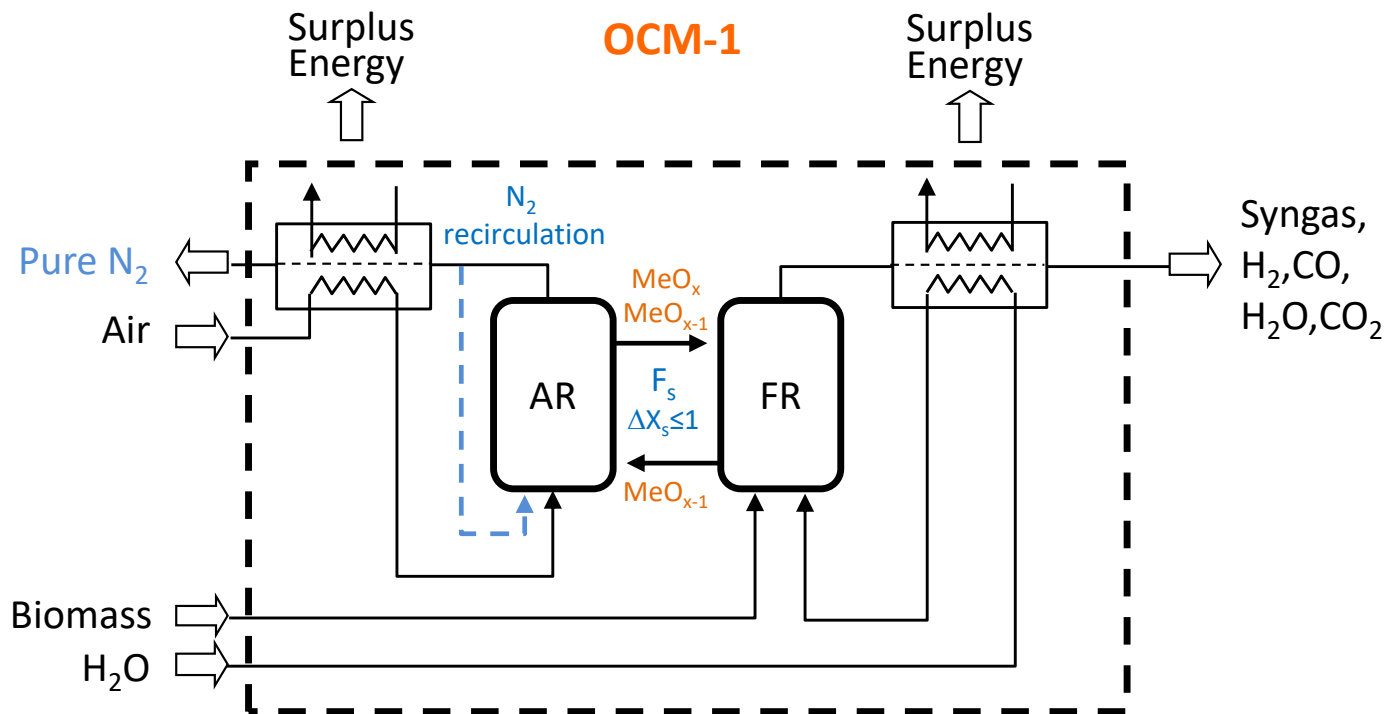
Temperature, T

Control of the oxygen in BCLG



- Controlled by the oxygen carrier circulation flow.
- The OC circulation flow is fixed for a given oxygen-to-biomass ratio, λ
- The air-to-biomass ratio is higher to the oxygen-to-biomass ratio, λ

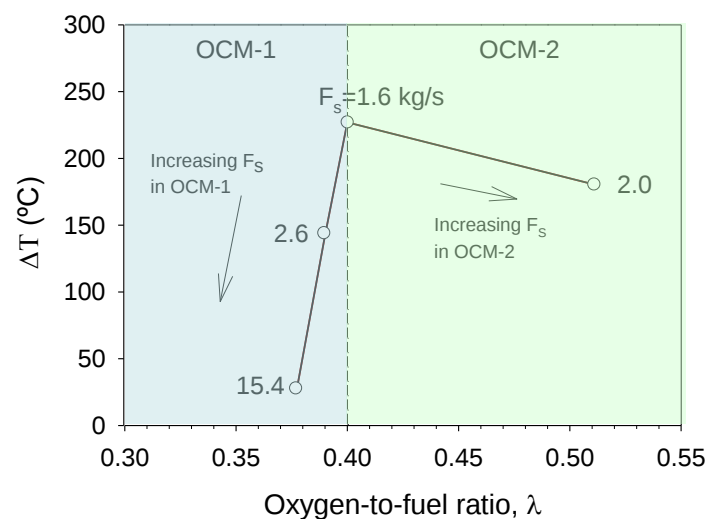
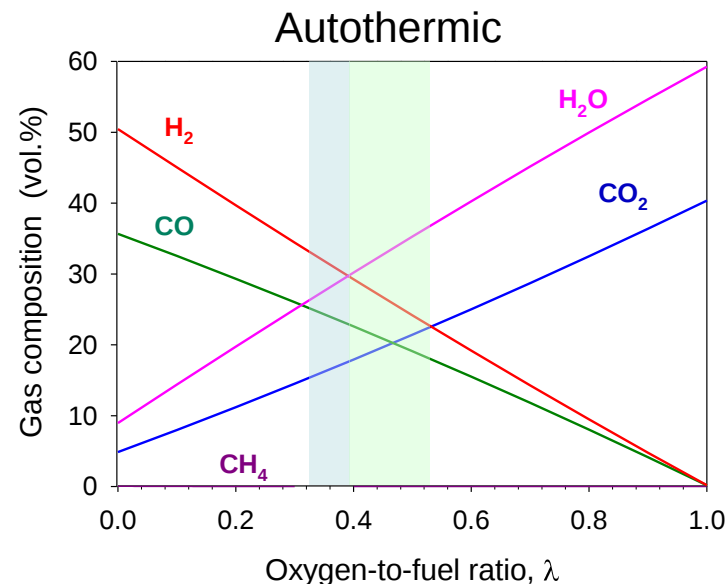
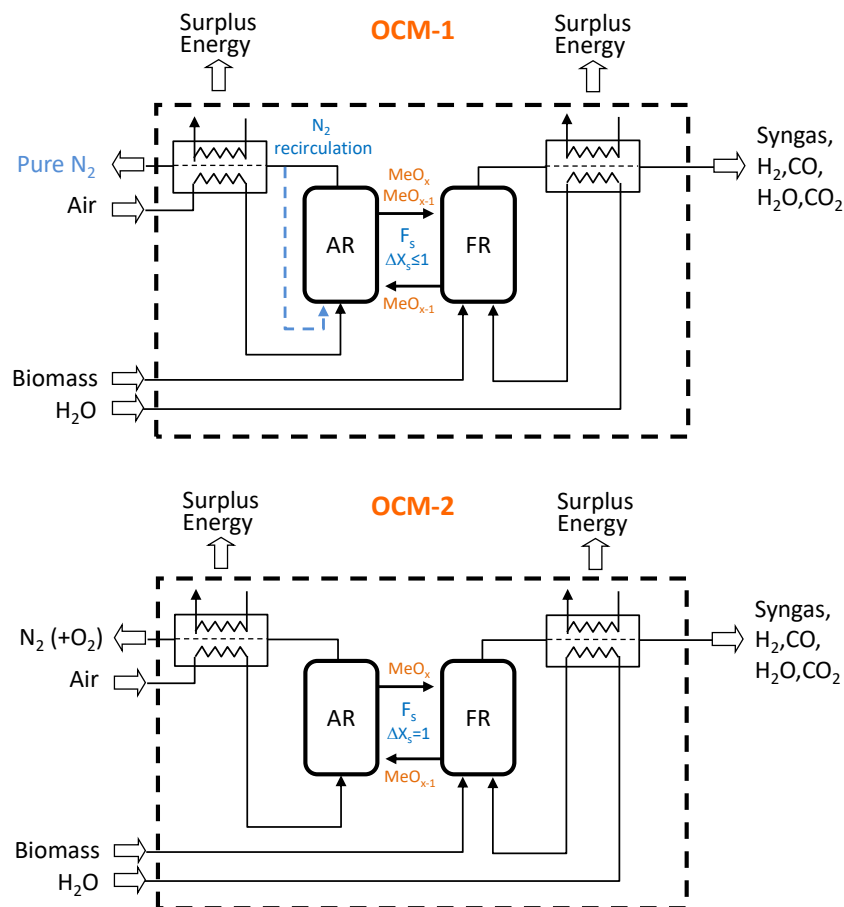
Control of the oxygen in BCLG



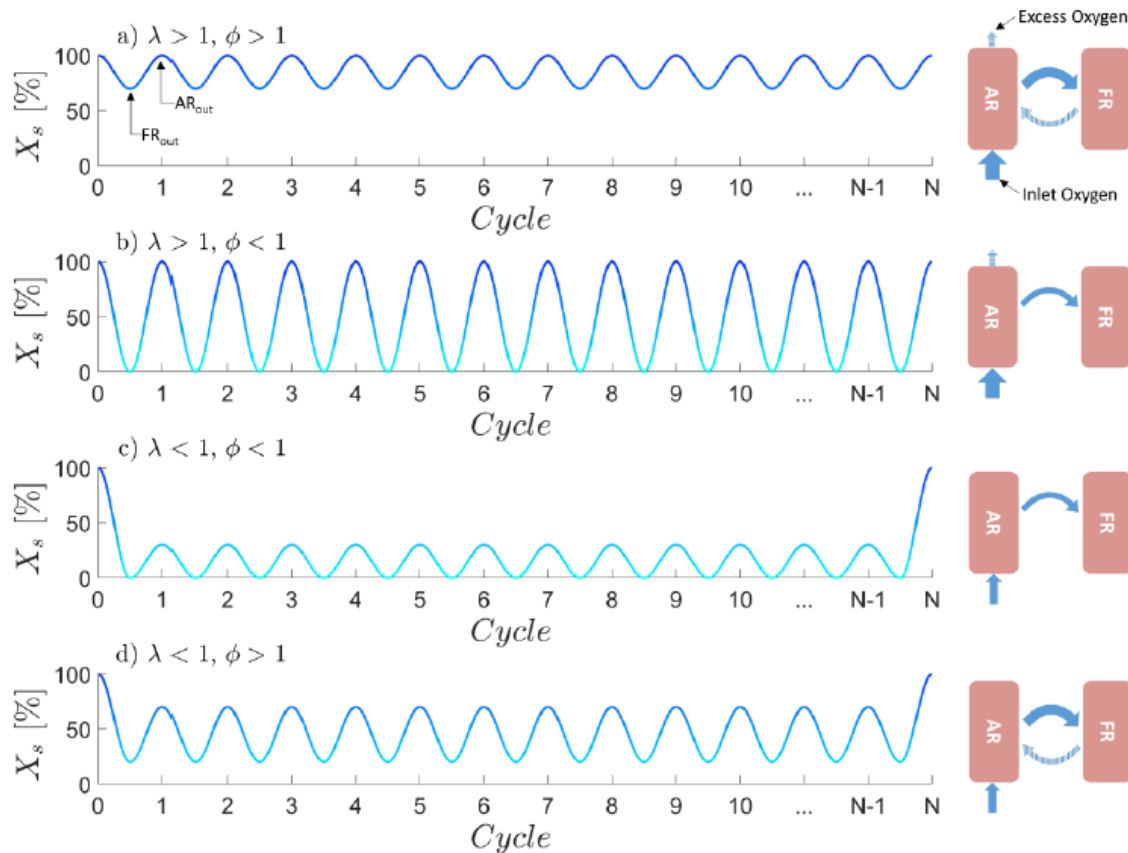
- Controlling the air-to-biomass ratio.
- Oxygen lattice transferred, λ , does not depend on the OC circulation flow
- Pure N_2 is obtained at the AR outlet

Control of the oxygen in BCLG

OCM-1 is preferred than OCM-2



Lines correspond to autothermal conditions



CLC

CLC

OC- Low oxygen transport capacity

CLG (OCM-1)

Limited O_2 in AR.
Low circulation rate

CLG (OCM-1)

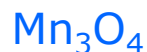
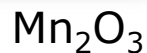
Limited O_2 in AR
High circulation rate

Figure 3. Different chemical looping modes (a–d) dependent on the air-to-fuel equivalence ratio λ and the oxygen-carrier-to-fuel equivalence ratio ϕ .

Oxygen carrier is highly reduced in FR and AR (OCM-1)

Redox pairs in BCLG

Mn-based



Mn

Fe-based



Fe

Ilmenite



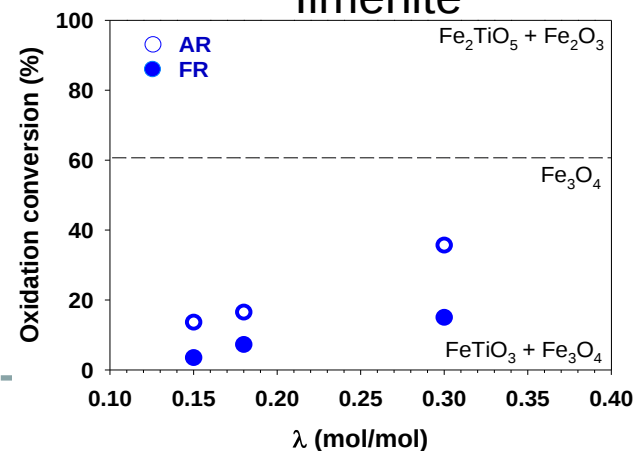
Fe. TiO_2

Fe-Al synthetic



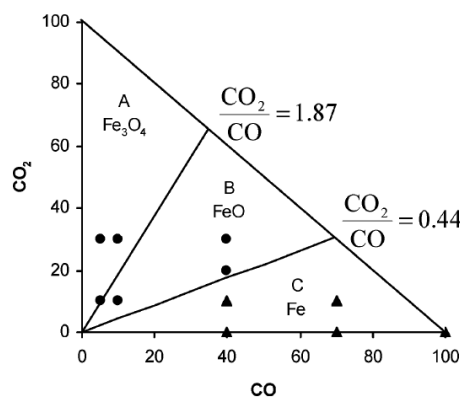
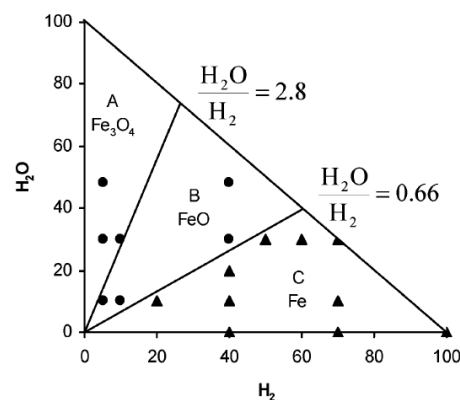
Fe. Al_2O_3

Ilmenite



Condori et al. Chem.Eng. J. 405, 2021, 126679

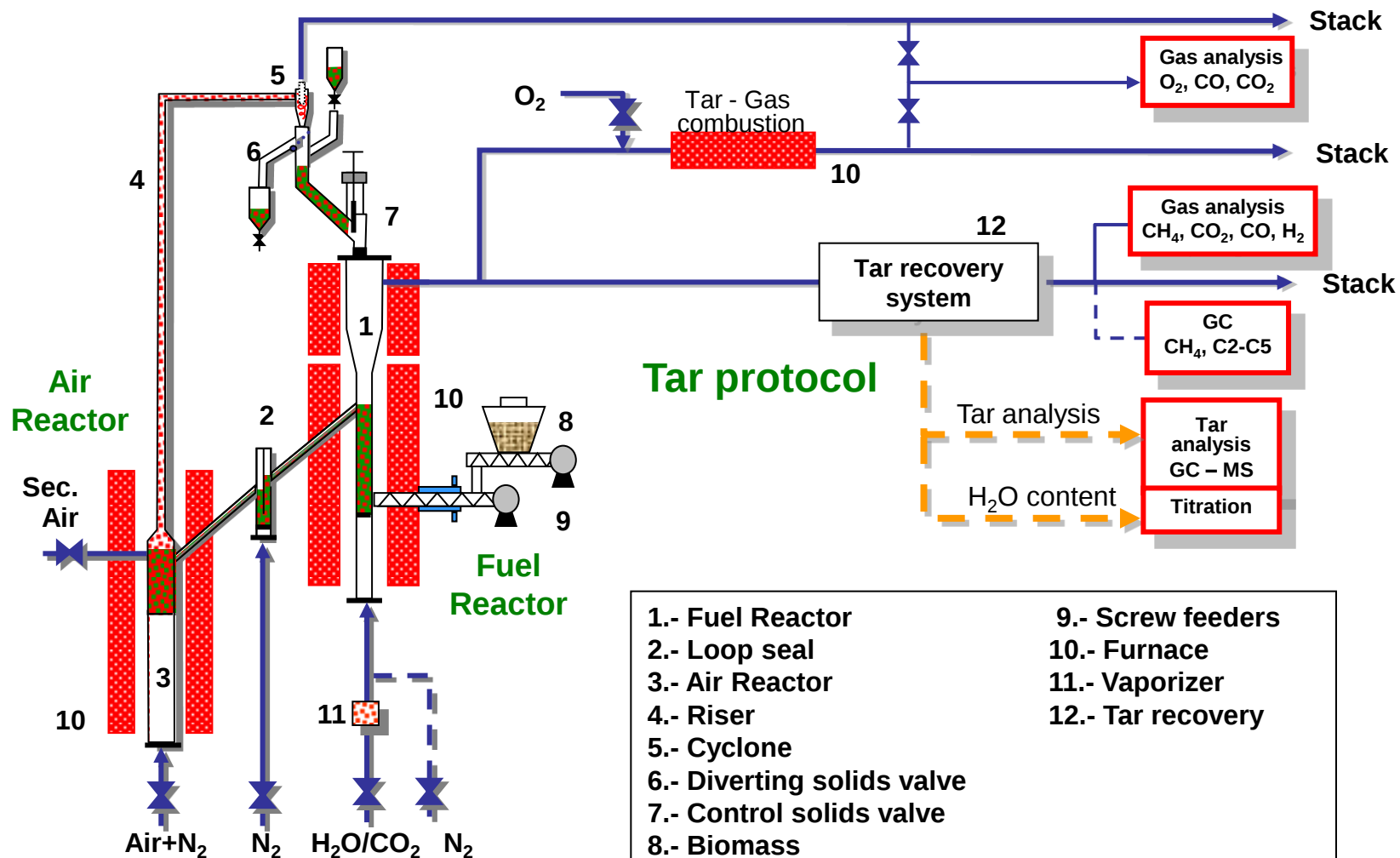
Thermodynamic data



Oxygen carrier
is highly reduced
in FR and AR (OCM-1)

Abad et al. Energy Fuels 21, 2007, 1843-1853.

1.5 kW_{th} CLG unit at ICB-CSIC



Summary of operating conditions at ICB-CSIC.

Unit	Oxygen carrier	Biomass	Temperature °C	Steam-to-fuel ratio kg/kg _{drybiomass}	Oxygen-to-fuel ratio (λ) mol/mol	Gasification time at steady state h
1.5 kW_{th}	Ilmenite	IWP	820 - 940	0 - 0.9	0.15 - 0.35	55
	Fe ore	IWP	820 - 940	0 – 0.6	0.15 – 0.5	40
	LD slag	IWP, almond shells, olive stone	820 - 940	0 – 0.6	0.15 – 0.5	60
	Mn ore	IWP	820 - 940	0 – 0.6	0.15 – 0.5	50
	Fe10Al	Pine wood	820 - 940	0 – 0.6	0.2 – 0.4	34
	Fe20Al	Pine wood	820 - 940	0 – 0.6	0.2 – 0.6	38
	Fe25Al	Pine wood	820 - 940	0 – 0.6	0.2 – 0.4	31
50 kW_{th}	Ilmenite	Wheat Straw pellets with additives	820 -990	0.7	0.2 - 0.4	37

- >300 h of CLG operation under steady state conditions.
- No agglomeration observed in spite of the high reduction level of the OC

Summary of biomasses used at ICB-CSIC



Industrial
wood
pellets



Pine
wood



Olive
Stone



Almond
Shell



Wheat
straw
pellets +
additives

Woody biomass

Herbaceous or agricultural straw

S.V. Vassilev et al./Fuel 89 (2010) 913–933

931

WWB - wood and woody biomass (samples 1-28)
HAB - herbaceous and agricultural biomass (samples 29-72)
HAG - herbaceous and agricultural grass (samples 29-38)
HAS - herbaceous and agricultural straw (samples 39-47)
HAR - herbaceous and agricultural residue (samples 48-72)
AB - animal biomass (samples 73-74)
MB - mixture of biomass (samples 75-78)
CB - contaminated biomass (samples 79-86)
AVB - all varieties of biomass (samples 1-86)
P - peat
L - lignite
S - sub-bituminous coal
B - bituminous coal
A - algae

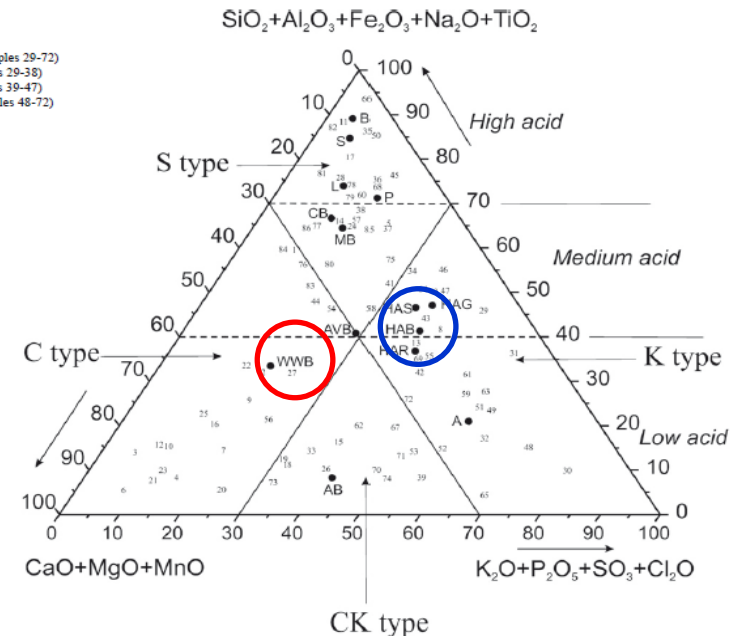


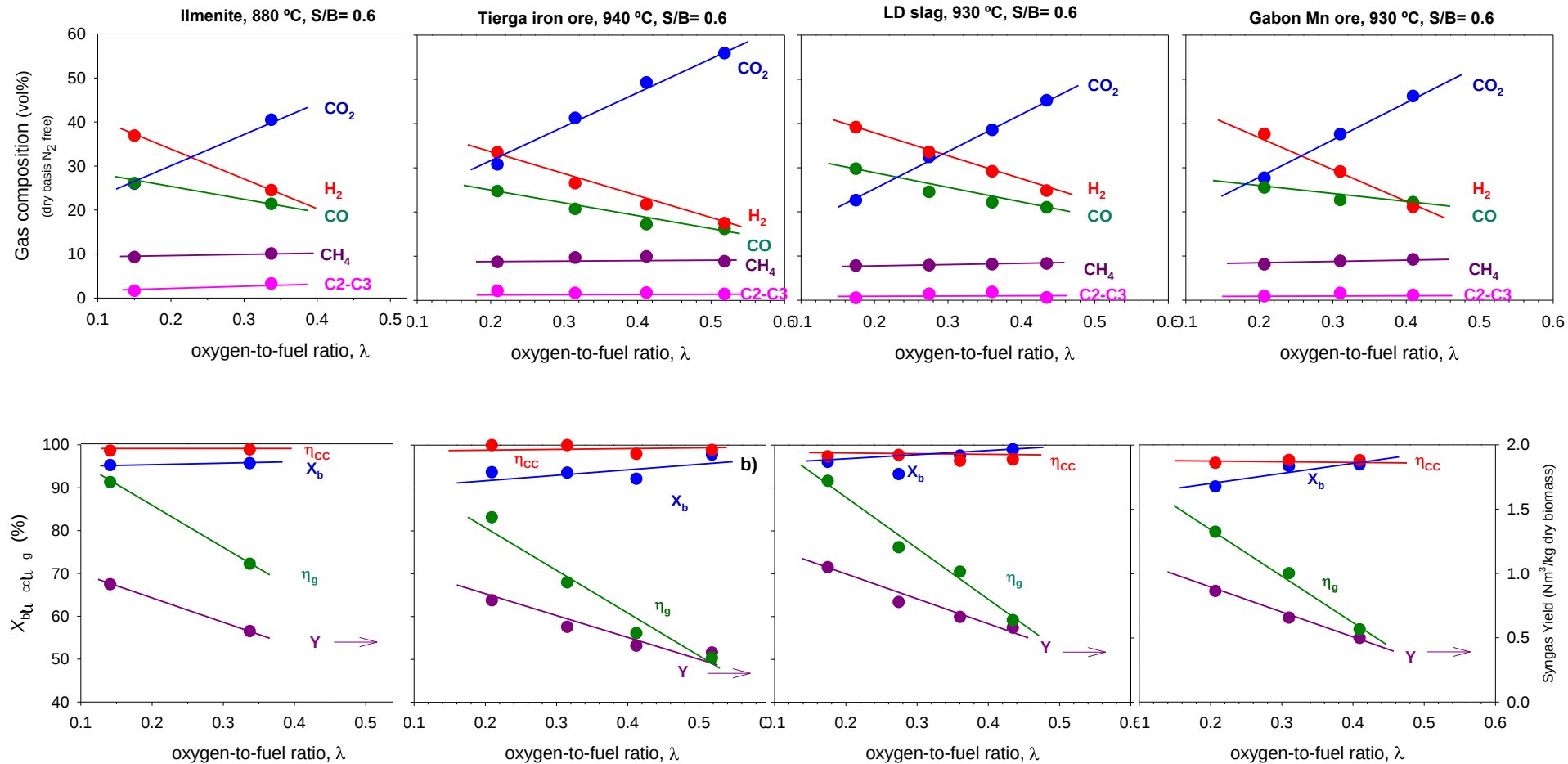
Fig. 4. Chemical classification system of the inorganic matter in high-temperature biomass ashes based on 78 variables. *Reference: [18]*

Ilmenite

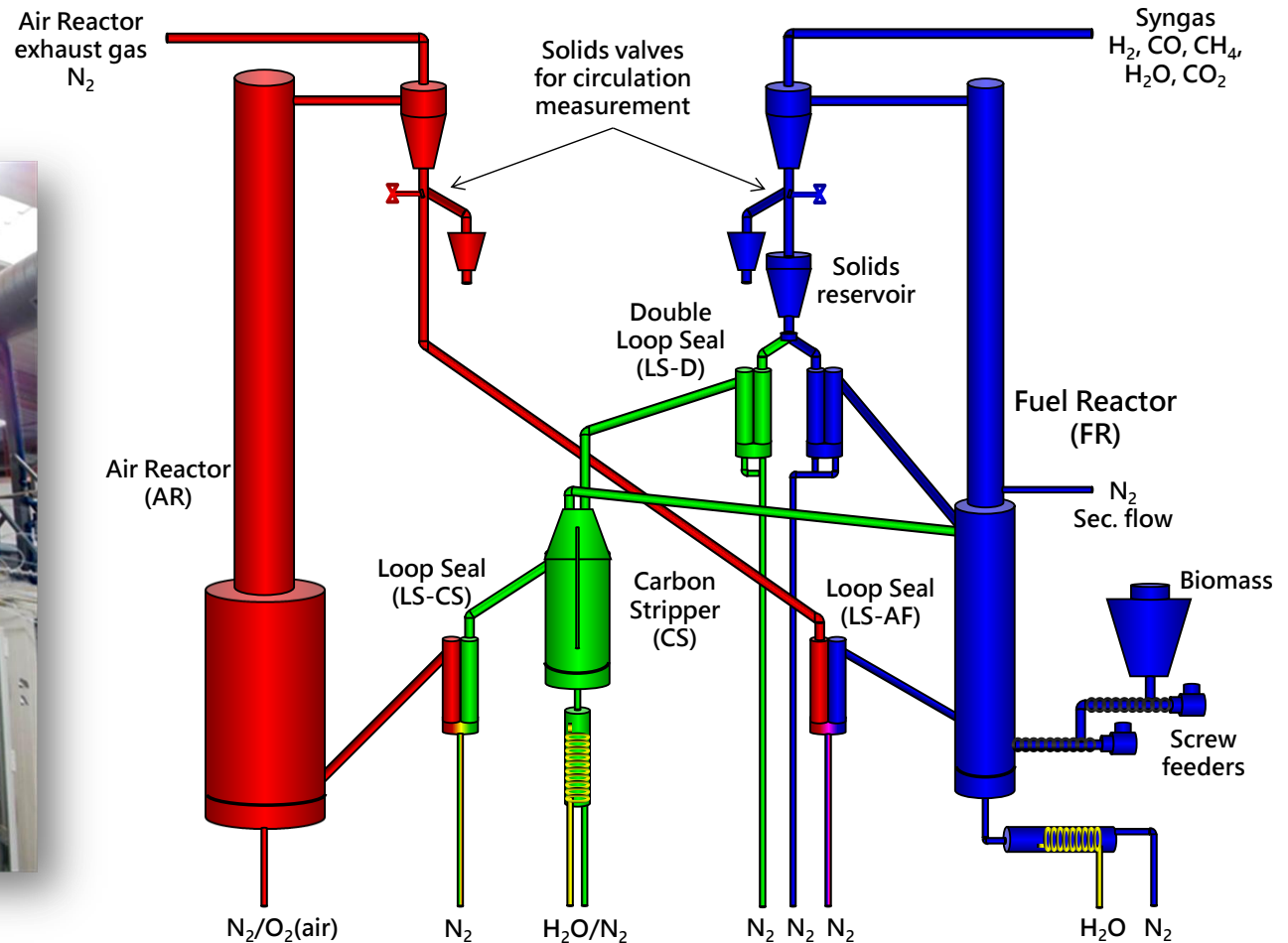
Tierga Fe ore

LD slag

Gabon Mn ore



50 kW_{th} Chemical Looping Gasification unit at CSIC



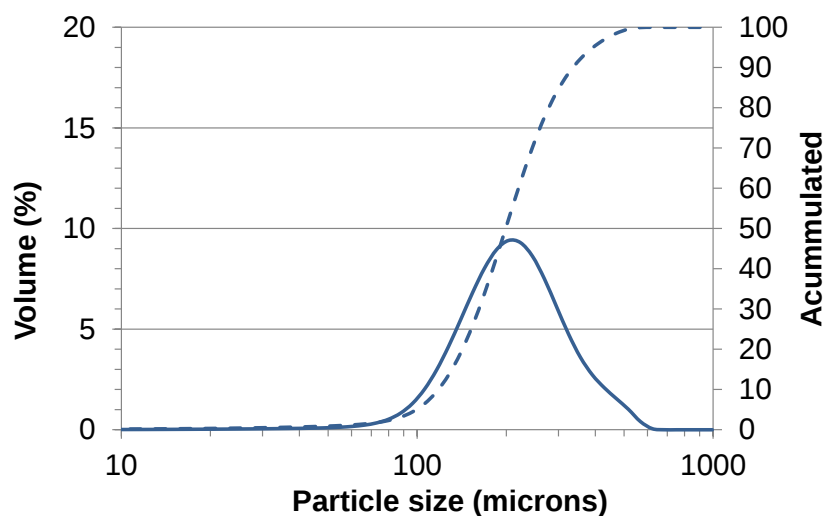
Control of the process by limiting the oxygen fed in Air Reactor

Oxygen-carrier

ILMENITE



Fresh ilmenite from Titania, Norway:
Calcination: 950 °C in air for 2 h
Circulated 6 h at 800 °C in air



Mean	10%	25%	50%	75%	90%
μm	μm	μm	μm	μm	μm
224	<123	<159	<209	<273	<353

active phases	
Oxidized	Reduced
Fe_2TiO_5	FeTiO_3
Fe_2O_3	Fe_3O_4

Calcined ilmenite

Fe_2TiO_5	54.7 wt. %
Fe_2O_3	11.2 wt. %
TiO_2	28.6 wt. %
Inerts	5.5 wt. %

PROPERTIES	Calcined ilmenite
Particle size (μm)	100-300
True density (kg/m^3)	4100
$R_{O,ilm}$ (%)	4.0
Porosity (%)	1.2
Crushing strength(N)	2.2
XRD (main species)	Fe_2TiO_5 , Fe_2O_3 , TiO_2



Proximate analysis

Moisture content	wt.%; wb	10.3
Ash content	wt.%; db	7.0
Volatile matter	wt.%; db	77.7
Fixed carbon ⁽¹⁾	wt.%; db	15.3

Ultimate analysis

C	wt.%; db	46.1
	wt.%; daf	49.6
H	wt.%; db	5.8
	wt.%; daf	6.2
N	wt.%; db	0.39
	wt.%; daf	0.42
S	wt.%; db	0.105
	wt.%; daf	0.113
Cl	wt.%; db	0.100
	wt.%; daf	0.108

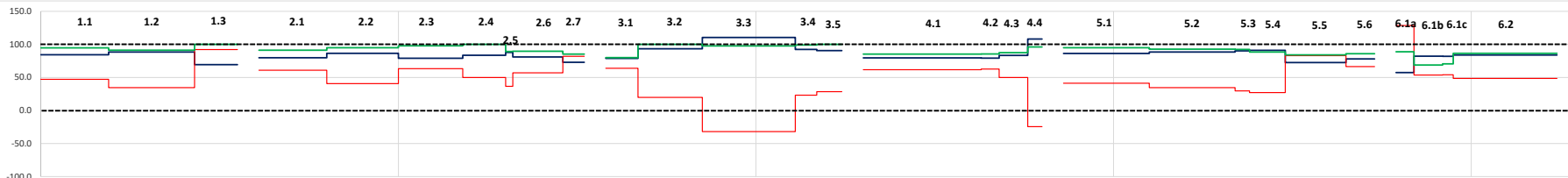
Net calorific value

NCV	MJ/kg; db	17.2
	MJ/kg; daf	18.5

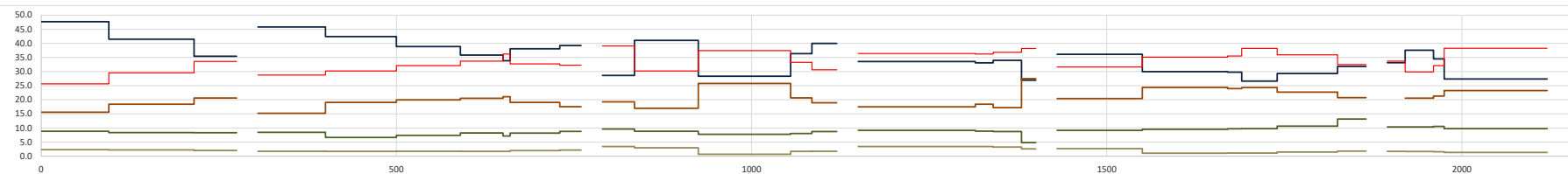
Major elemental ash composition in additive wheat straw pellets

Al	mg/kg (w/w)	158
Ca	mg/kg (w/w)	11,500
Fe	mg/kg (w/w)	172
K	mg/kg (w/w)	7,330
Mg	mg/kg (w/w)	982
Na	mg/kg (w/w)	186
P	mg/kg (w/w)	448
Si	mg/kg (w/w)	10,200
Ti	mg/kg (w/w)	10

Relevant efficiency parameters



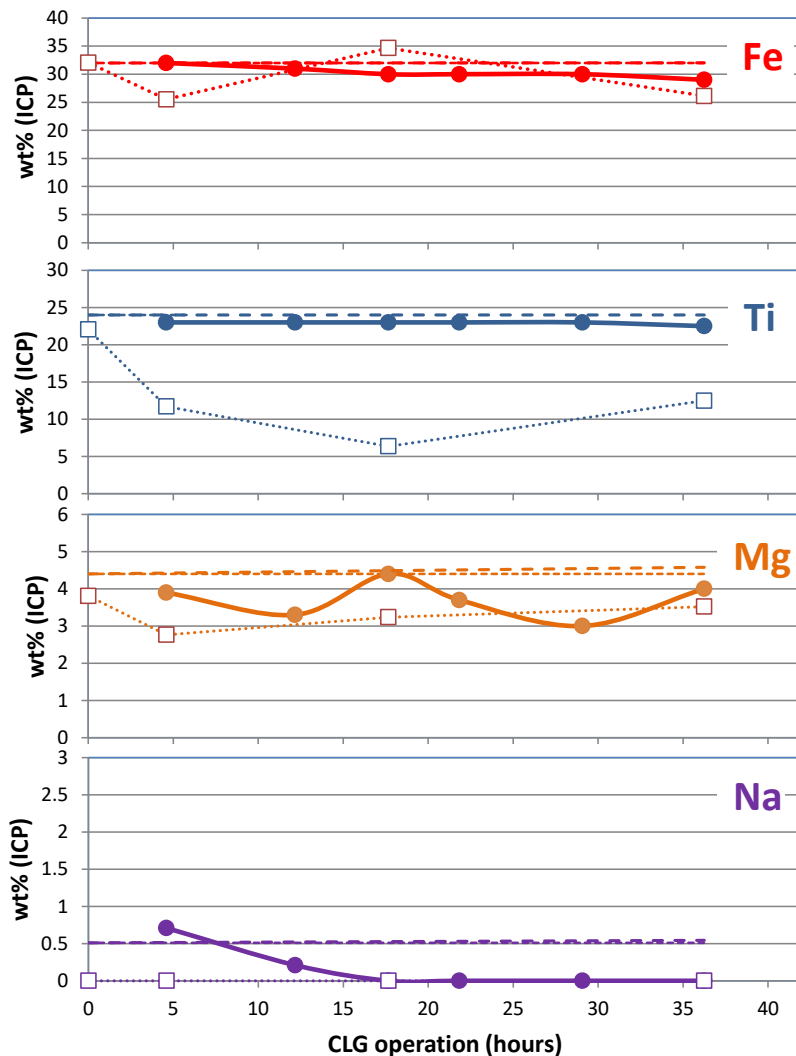
Gas composition



Main Achievements

- 6 days CLG. 37 h of continuous CLG operation
- 18 kW_t (4 kg/h of biomass)
- Smooth operation
- Several operating conditions analyzed
 - Temperature, $T = 800 - 980\text{ }^{\circ}\text{C}$
 - Oxygen to fuel ratio, $\lambda = 0.13 - 0.4$
 - Steam to biomass ratio, $S/B = 0.7$
- Tar recovery
- **Agglomeration analyzed**

Evolution of components (ICP)

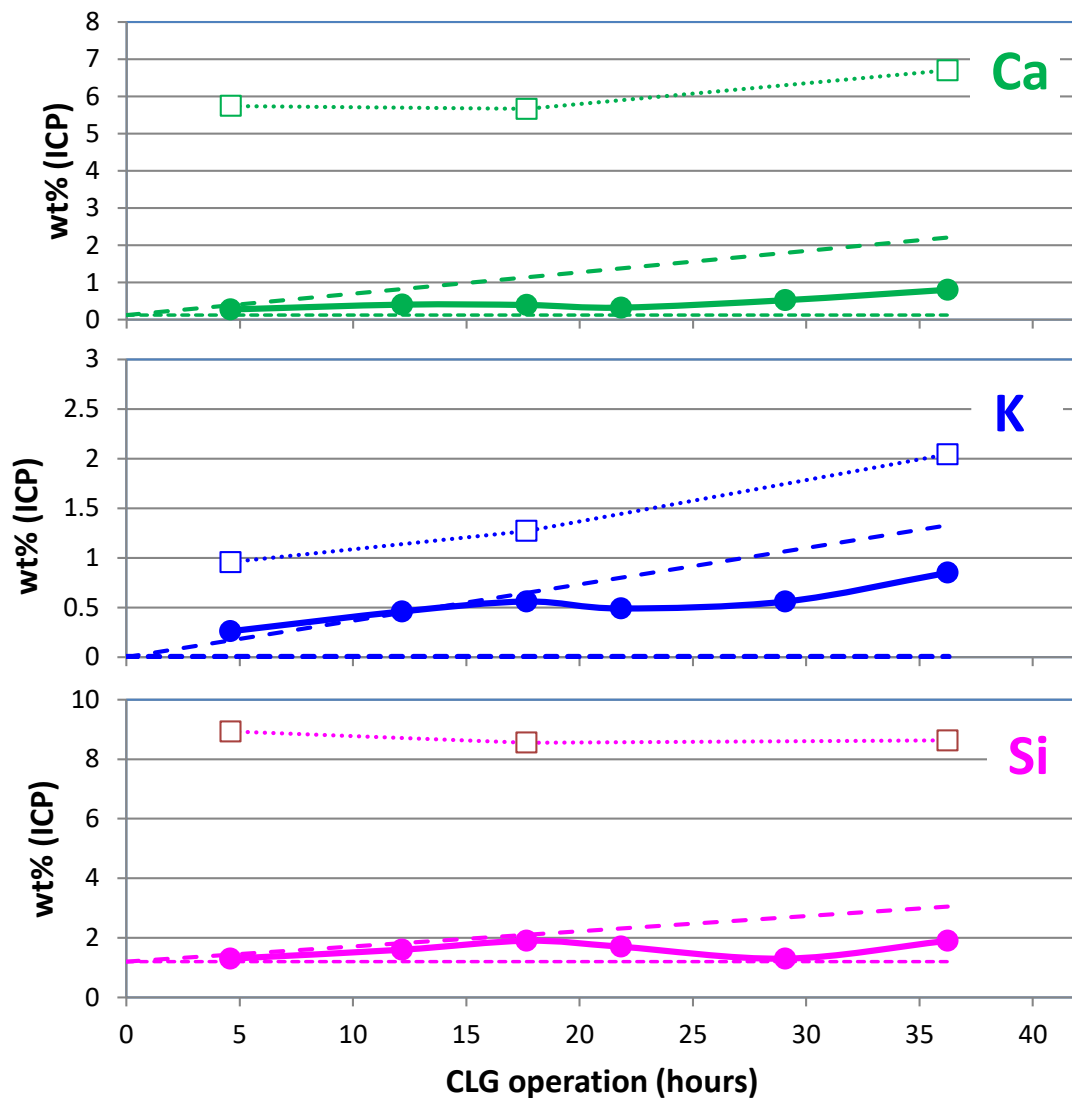


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Dot line: already present in Ilmenite
 Discontinuous line: ILM + fed in biomass
 Dots: ICP content in OC
 Square: ICP content in AR fines

80 kg oxygen carrier
 145 kg biomass
 37 hours CLG



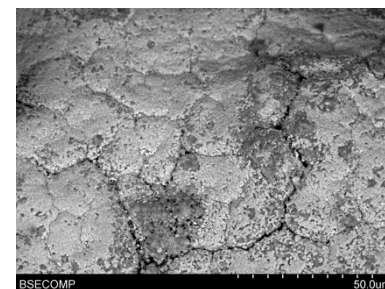
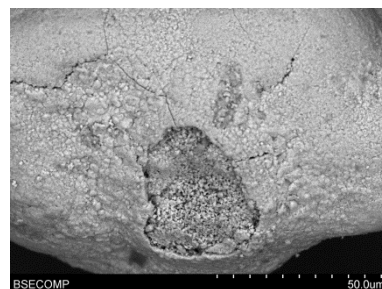
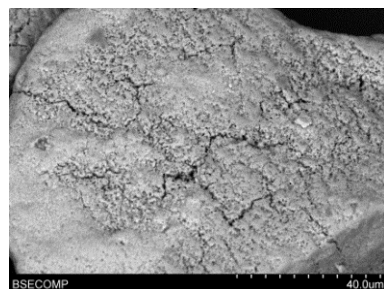
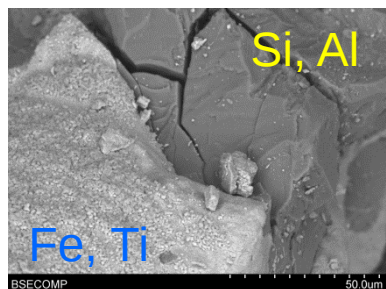
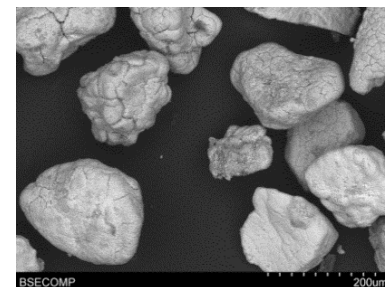
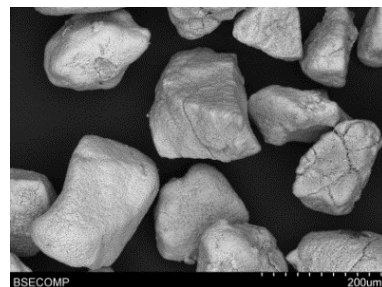
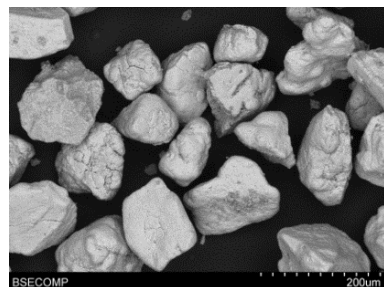
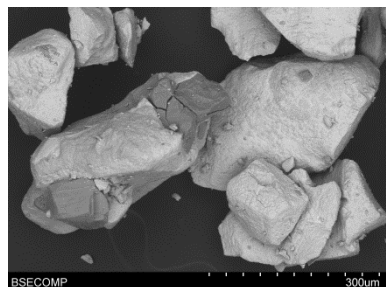
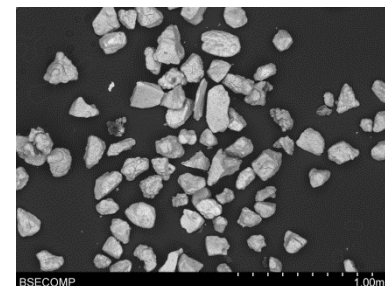
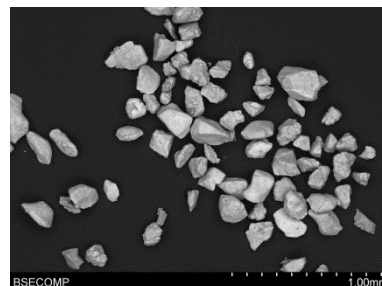
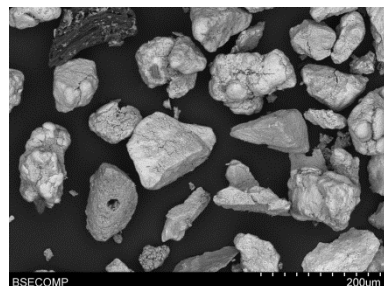
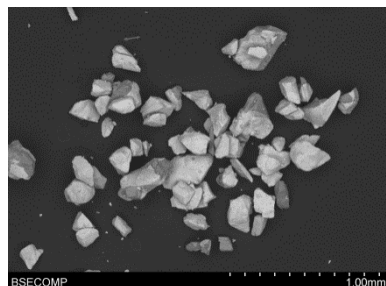
Dot line: already present in Ilmenite
 Discontinuous line: ILM + fed in biomass
 Dots: ICP content in OC
 Square: ICP content in AR fines

Calcined

Day 1 CLG

Day 3 CLG

Day 6 CLG



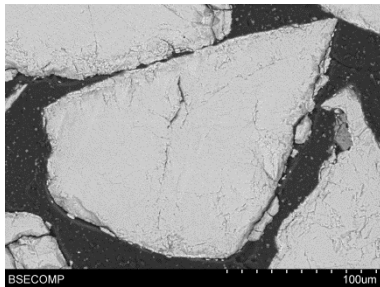
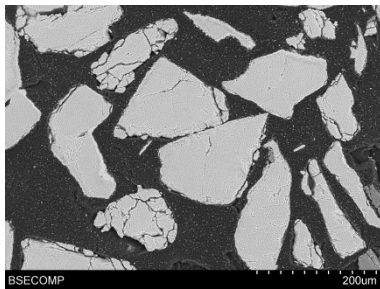
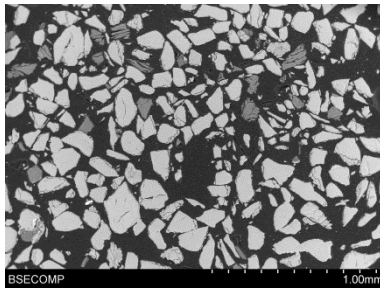
Particles with edges
Fines stuck
White particles: Fe, Ti
Black particles: Si, Al

Rounded particles
Some cracks

Some particles with
detached shell

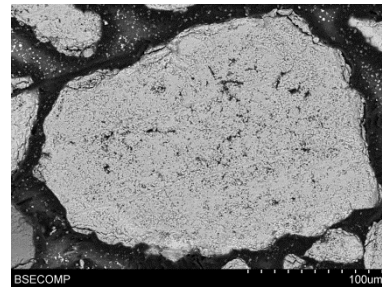
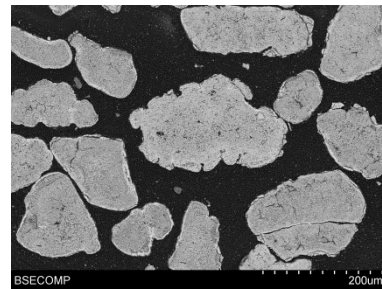
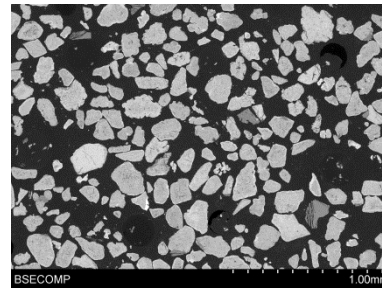
Good appearance
No agglomeration

Calcined



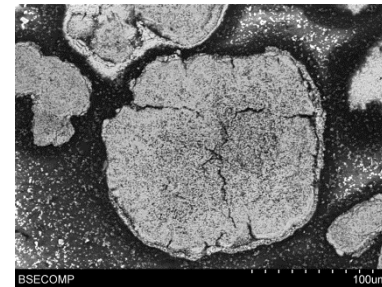
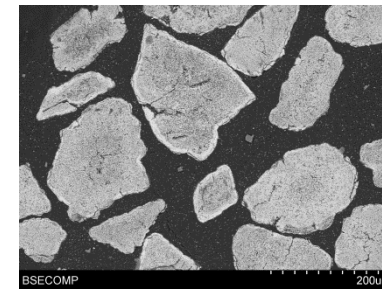
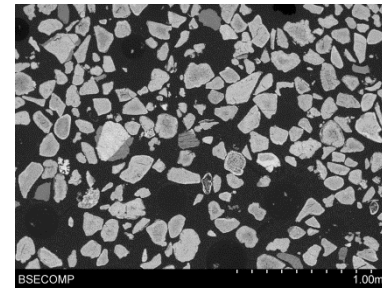
Particles with edges
Some internal cracks

Day 1 CLG



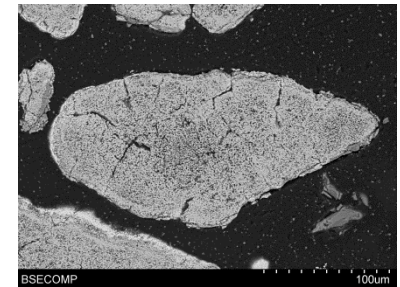
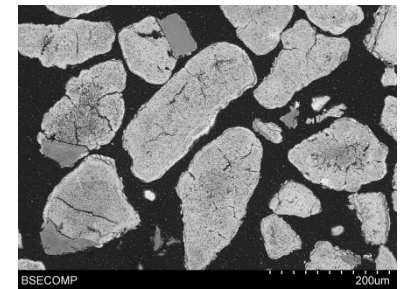
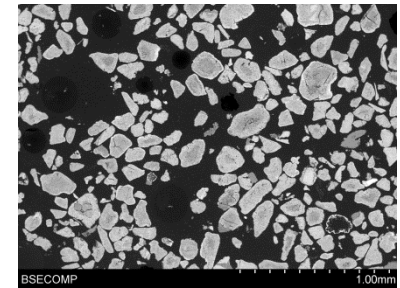
Rounded particles
White and black particles
External layer

Day 3 CLG



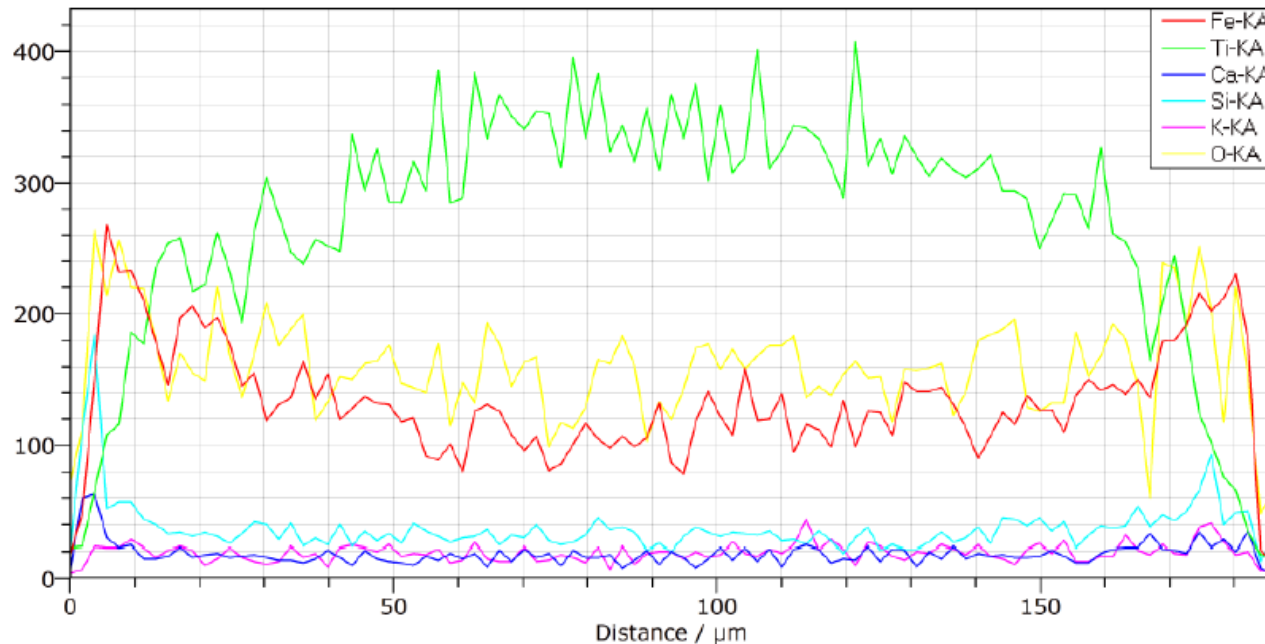
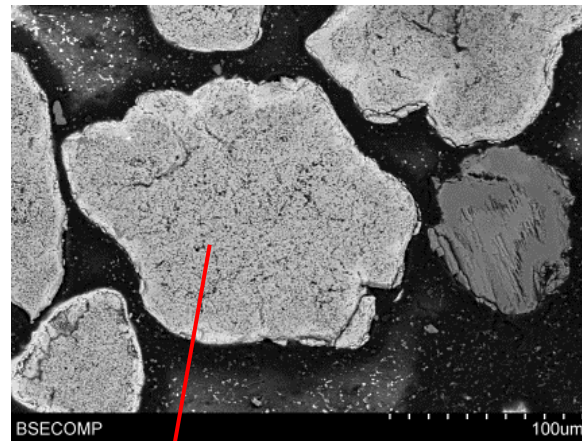
External layer
Some internal cracks

Day 6 CLG

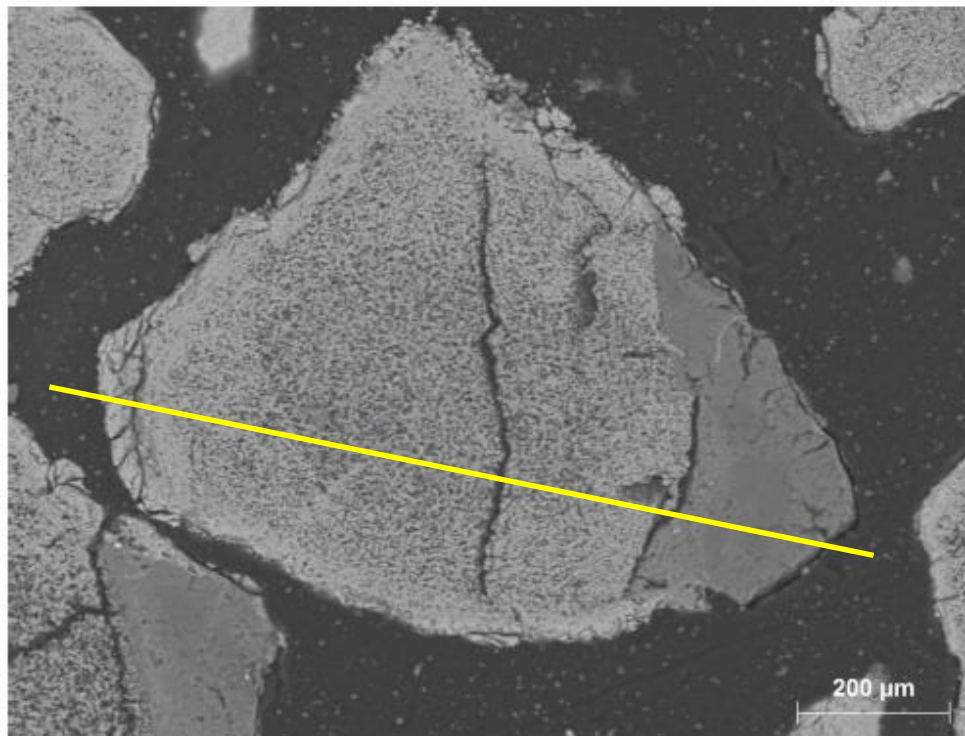


Good appearance
No agglomeration

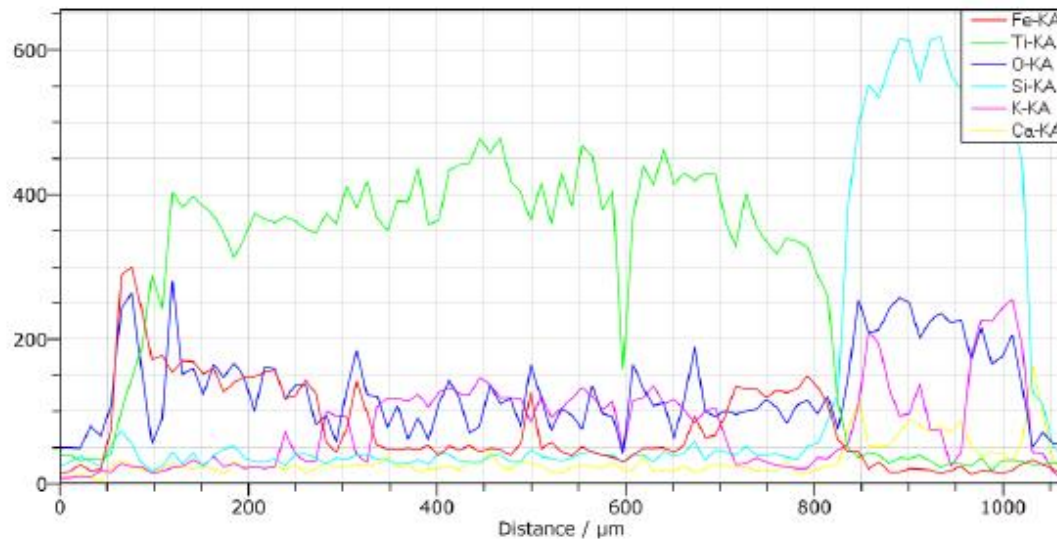
Fe Ti Ca Si K O



Fe migration towards the external surface of the particles (Fe oxide)



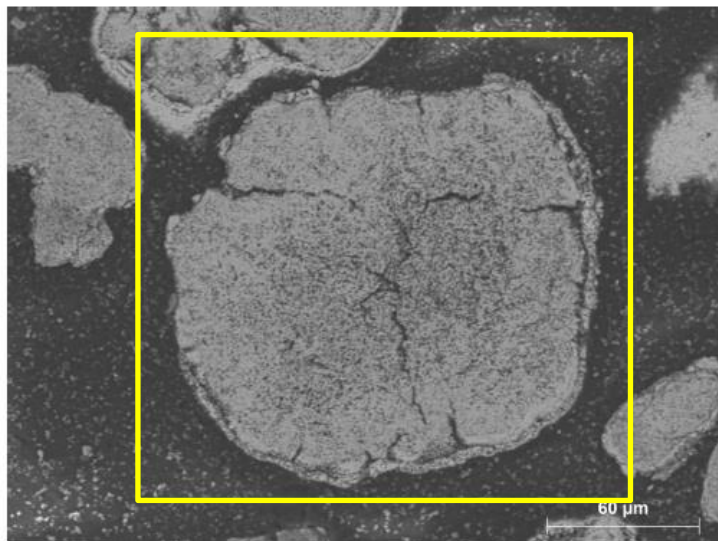
K goes to Si-enriched locations



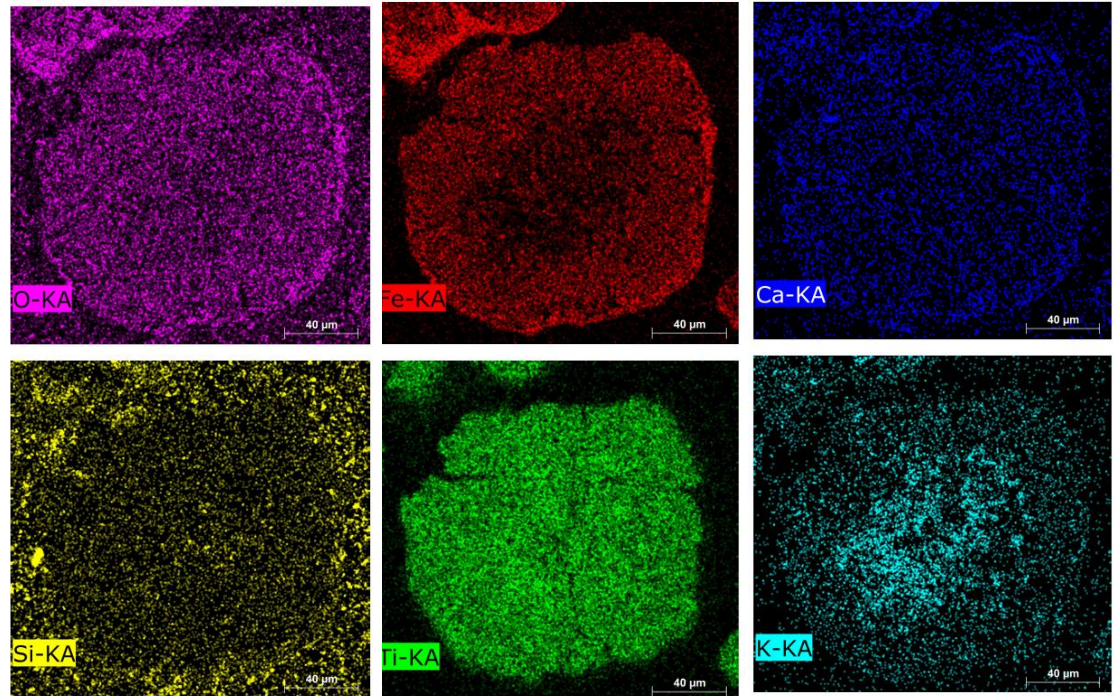
Fe Ti Ca Si K O

Looking for K, Ca and Si

- Some migration of Fe occurs towards the external surface of particle in form of Fe oxide (no Titania).
- K penetrate inside the particles when cracks are present.
- Si and Ca are preferently deposited in the external layer of the particle.



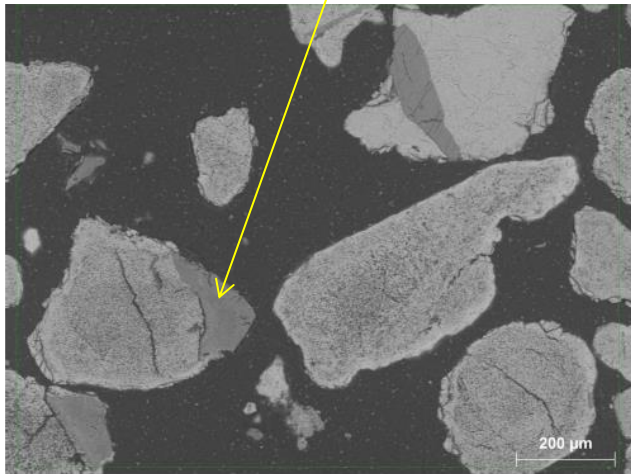
Mapping IlmFRD4-2-x450



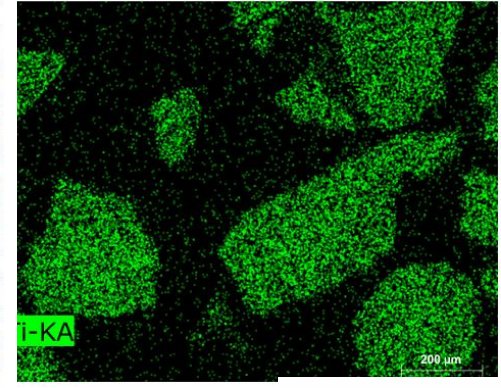
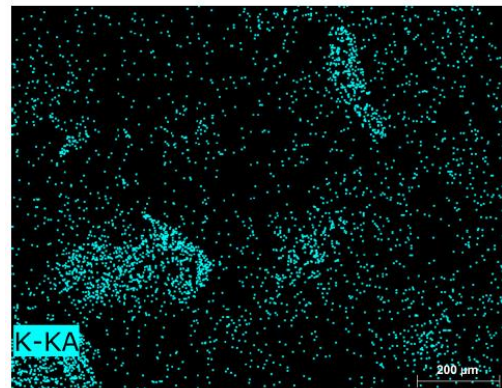
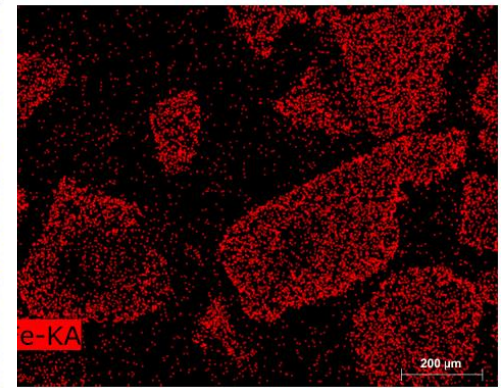
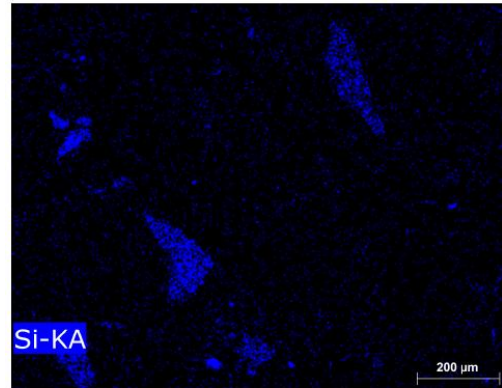
Day 6 CLG-AR

- Fe migration outwards
- K is enriched in zones where Si is present (even if it is inside the particles).

Dark zone is mainly Si,Al-Mg

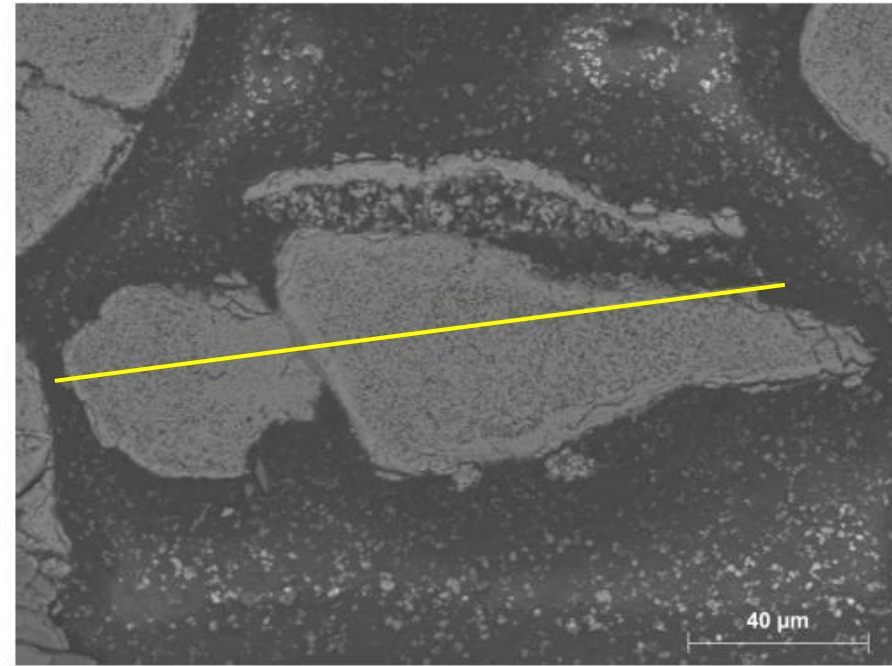


Mapping IlmARD6-6-x230

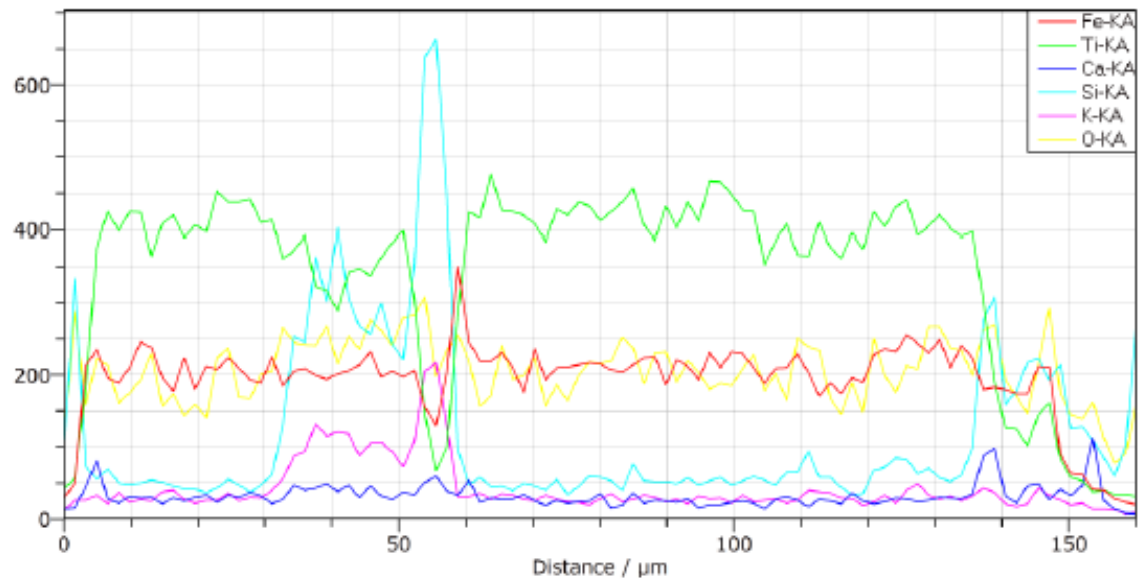


Looking for agglomeration, looking for K

- Particles after 37 h under CLG operation behaved well and no operational problems were observed.
- SEM-EDX studies showed separate particles.
- **No agglomeration** was produced.
- In the whole study, only 1 small agglomeration case was observed in day 4 in a particle.



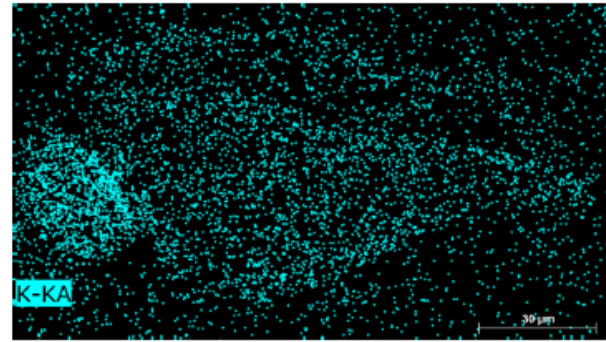
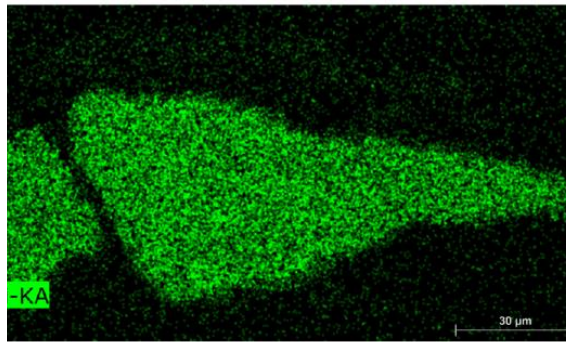
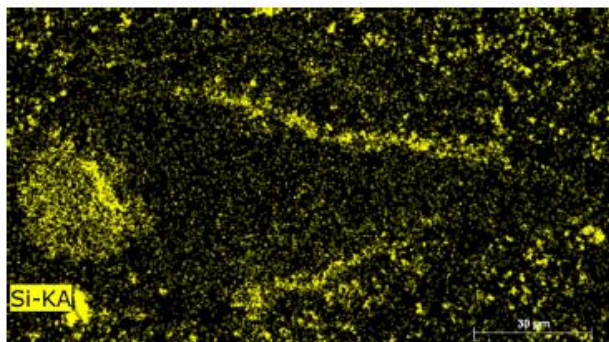
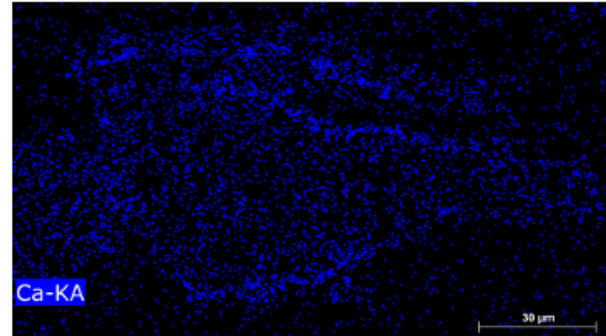
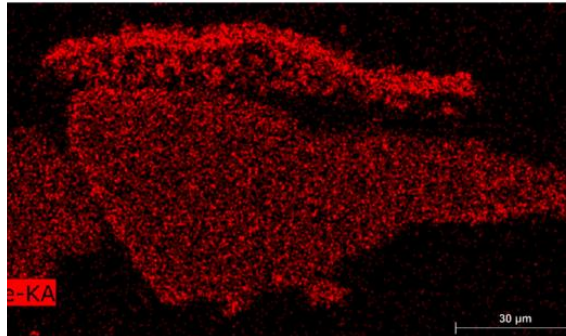
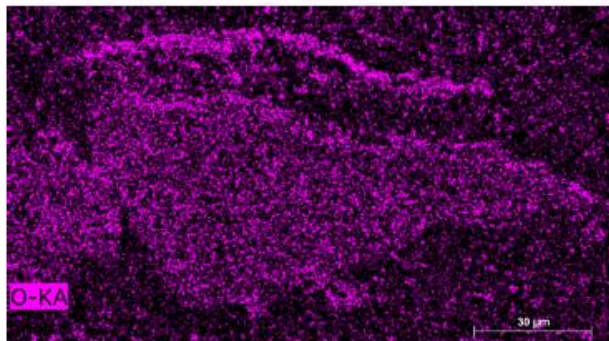
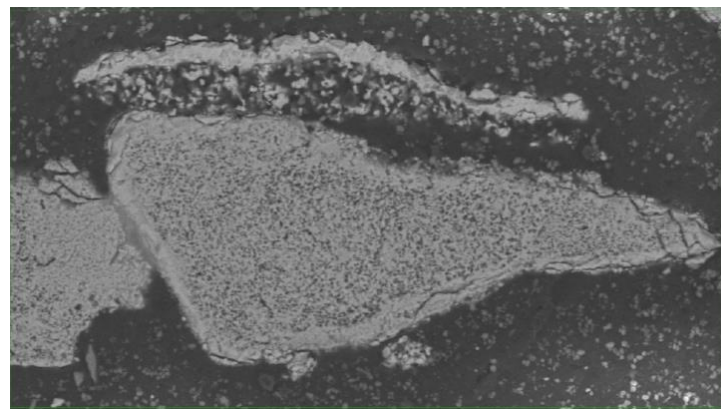
Fe Ti Ca Si K O



Linescan IImFRD4-4-x650
Mapping IImFRD4-3-x800

Looking for agglomeration, looking for K

- The shell detached is Fe-O (no Ti)
- K concentrates preferably in Si sites
- Ca and Si seems to be concentrated in the external surface of the particle



Summary (from 50 kW_{th} unit tests)

- **NO agglomeration** was detected during 37 h continuous CLG operation using wheat straw pellets with Ca additive.
- Some migration of Fe outwards the particle surface.
- Some layers detached from particles, but not so relevant as expected.
- Ca, Si and K present in biomass is partially deposited in the oxygen carrier.
- Ca and Si are deposited preferably in the external surface of particles.
- K appeared preferably in Si locations already present on ilmenite.
- K also appeared in internal zones of ilmenite

6th
International
Conference on
Chemical looping
2022

Hotel Hiberus. Zaragoza (Spain), September 19-22, 2022



<http://chemical-looping2020.com/>

~~Zaragoza (Spain), September 21-24~~

Set the new dates

Zaragoza (Spain), September 19-22, 2022

Topics of interest:

- Chemical looping combustion
- Chemical looping gasification/reforming
- Chemical looping oxygen carriers
- Chemical looping reactor design and operation
- Novel chemical looping processes
- Carbonate looping technology
- Modeling of chemical looping process
- Pilot plants and scale-up

We are looking forward seeing you in Zaragoza in 2022

Prof. Juan Adánez
Chairman of the Conference

CSIC funding in BCLG

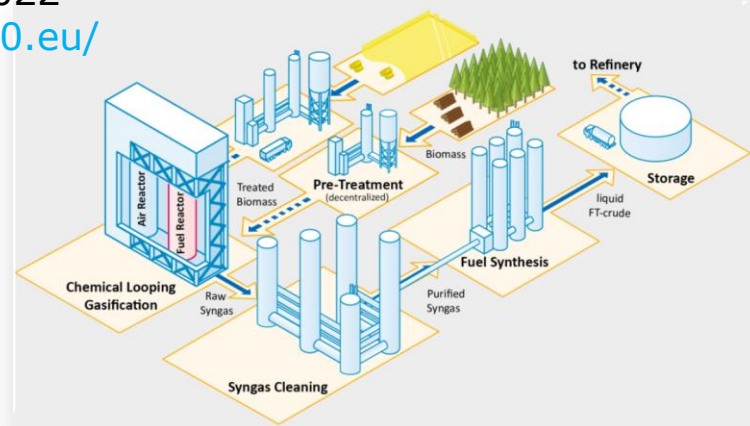


H2020-No. 817841

Chemical Looping Gasification for Sustainable Production of Biofuels

Nov 2018 – Nov 2022

<https://clara-h2020.eu/>



ENE2017-89473-R



Gasificación de biomasa mediante tecnologías de *Chemical Looping* para producción de gas de síntesis/H₂ con captura de CO₂

Jan 2018 – Sep 2021

Combustion and Gasification Group

Instituto de Carboquímica (ICB) Zaragoza
Consejo Superior de Investigaciones Científicas (CSIC)

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Research on
Chemical Looping
since 2000



Many Thanks. Questions?