

High volatiles conversion in a dual stage fuel reactor system for Chemical Looping Combustion of wood biomass

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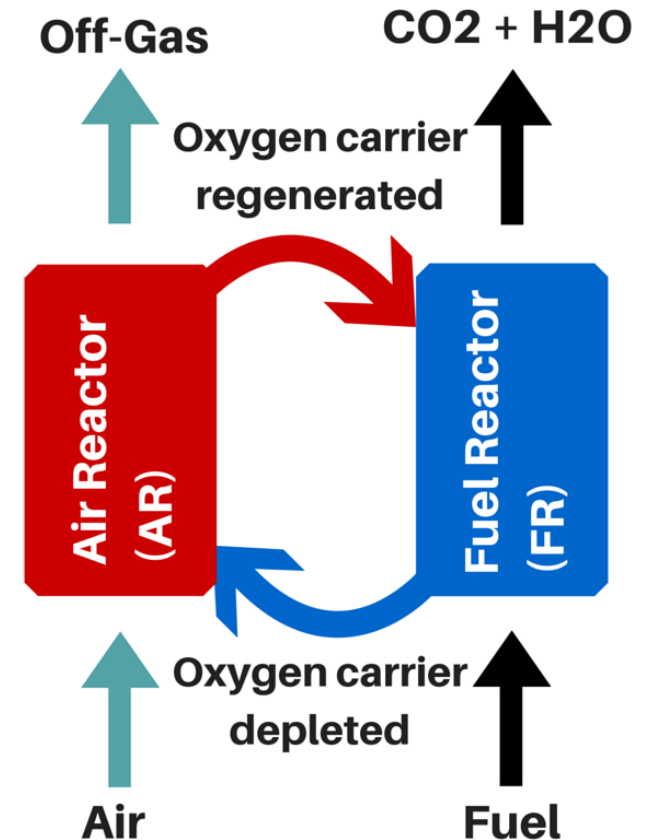
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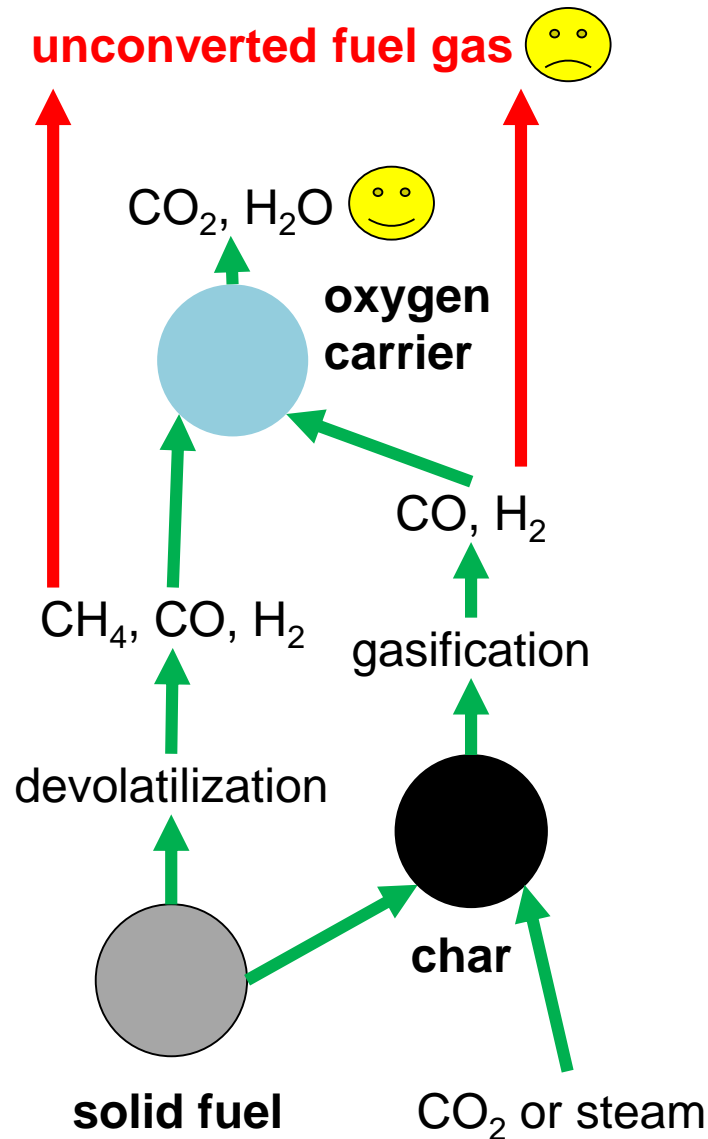
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Chemical Looping Combustion

- Carbon capture process
- Solid oxygen carrier (OC) is oxidized/reduced (reactor-regenerator system)
- iG-CLC: solid fuel is gasified
- Complex heterogeneous reaction network
 - Oxygen carrier reactions
 - Solid fuel conversion
- Conversion kinetics differ from those of regular air firing





Solid fuel conversion pathway

In-situ Gasification Chemical Looping
Combustion

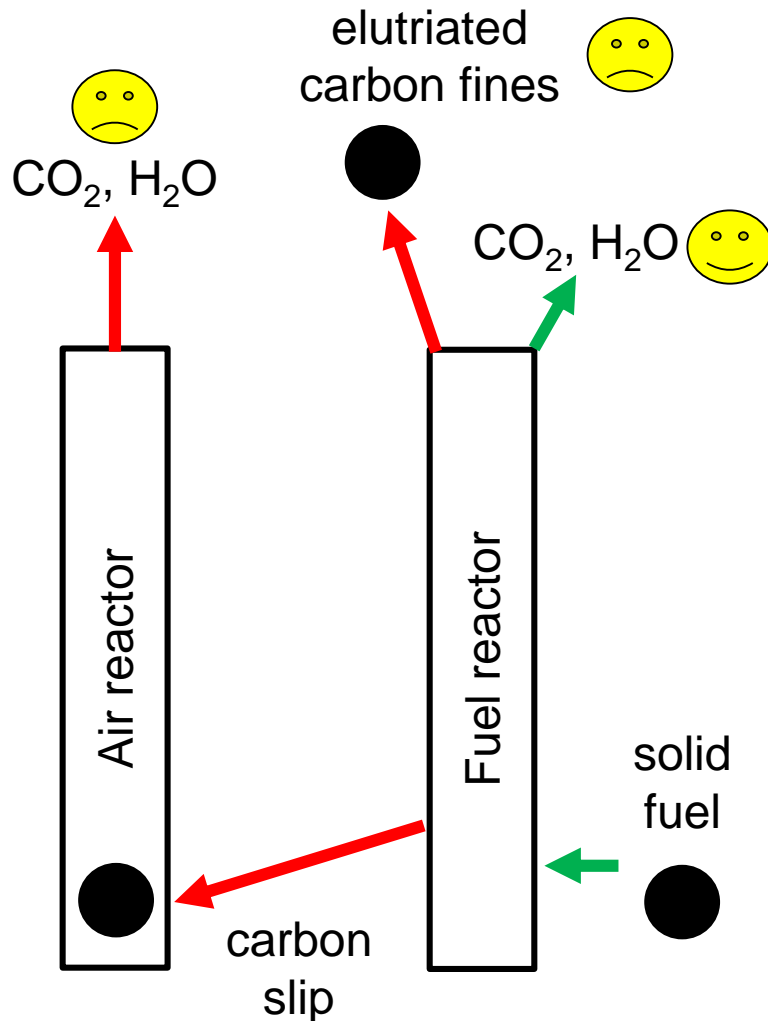
- Solid fuel releases volatiles
- Remaining char is gasified
- CO, H₂ and CH₄ can react with oxidized carrier

Main issue

- Parts of fuel gases do not contact or react with oxygen carrier (bypass)
- CO, H₂ and CH₄ leave system unconverted
→ plant efficiency ↓

Expressed by the term oxygen demand Ω_{OD} :

$$\Omega_{OD} = \frac{\text{oxygen needed for fuel gases leaving FR}}{\text{oxygen needed for total fuel conversion}}$$



Carbon slip with the oxygen carrier to AR

- CO₂ released in AR not captured
- Carbon capture η_{CC} :

$$\eta_{CC} = \frac{\text{carbonaceous gases (FR)}}{\text{carbonaceous gases (FR + AR)}}$$

Elutriated carbon fines

- Solid fuel not converted \rightarrow plant efficiency \downarrow
- Solid fuel conversion X_{sf} :

$$X_{sf} = 1 - \frac{\text{Carbon elutriated}}{\text{Carbon introduced}}$$

Strong fuel influence on performance

- Biomass has high volatile fraction
- Very reactive char
- **Mainly oxygen demand Ω_{OD} is an issue**



Investigation of general process behavior

- Fluid dynamics
- Chemical conversion

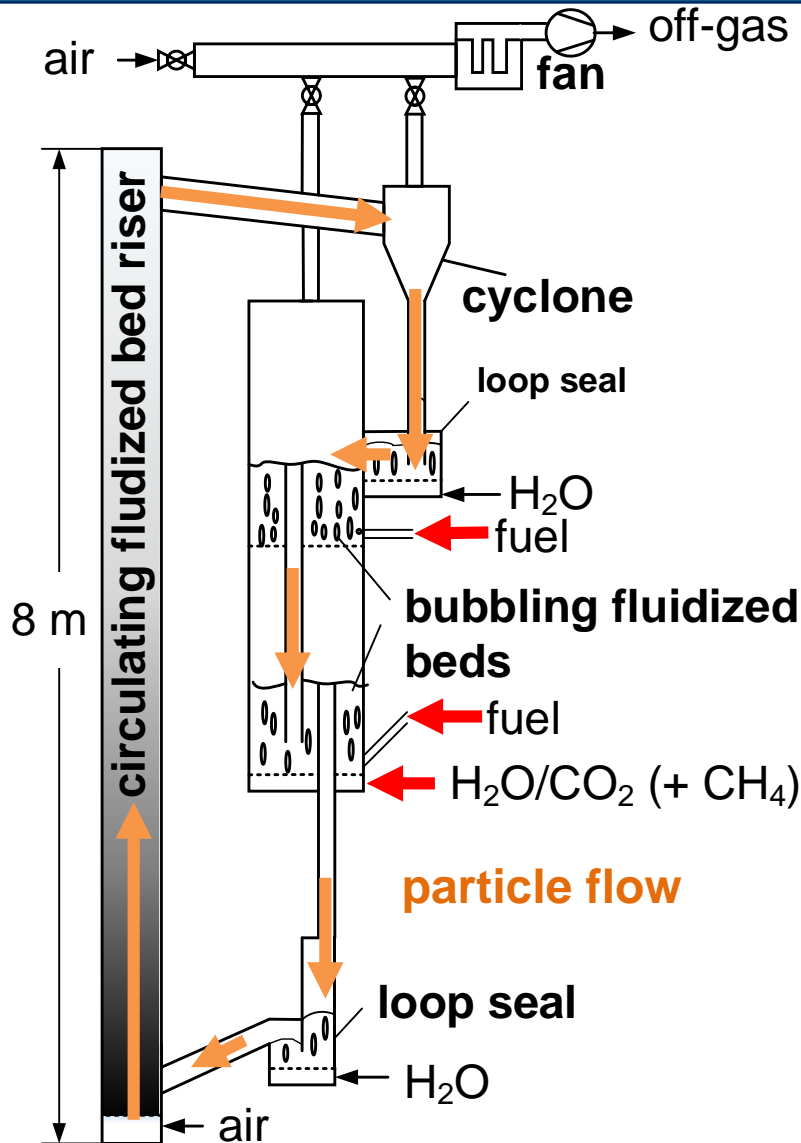
Experiments with various fuels

- Complex conversion behavior of lignite, bituminous coal and methane
- Interaction of fluid mechanics and chemical conversion (particle size)

Special focus on two-stage system

- Second stage for fuel gas conversion
- ✓ **Reducing oxygen demand**
- ✓ **Suitable design for high volatile fuels**
- Maybe suitable design for Biomass-CLC?**

[1] Haus et al., Analysis of a Two-Stage Fuel Reactor System for the Chemical-Looping Combustion of Lignite and Bituminous Coal, Energy Technology (2016)



Dimensions	height / m	diameter / m
air reactor	8	0.1
fuel reactor	4	0.25
bed height FR	0.6	per stage
cyclone	0.34	0.21
	width / m	length / m
siphon S1	0.16	0.13
siphon S2	0.13	0.25

Operation mode

- Air reactor -> fast fluidized bed (5 m/s)
- Fuel reactor -> bubbling bed (0.25 m/s)
- Siphons -> bubbling bed (0.2 m/s)

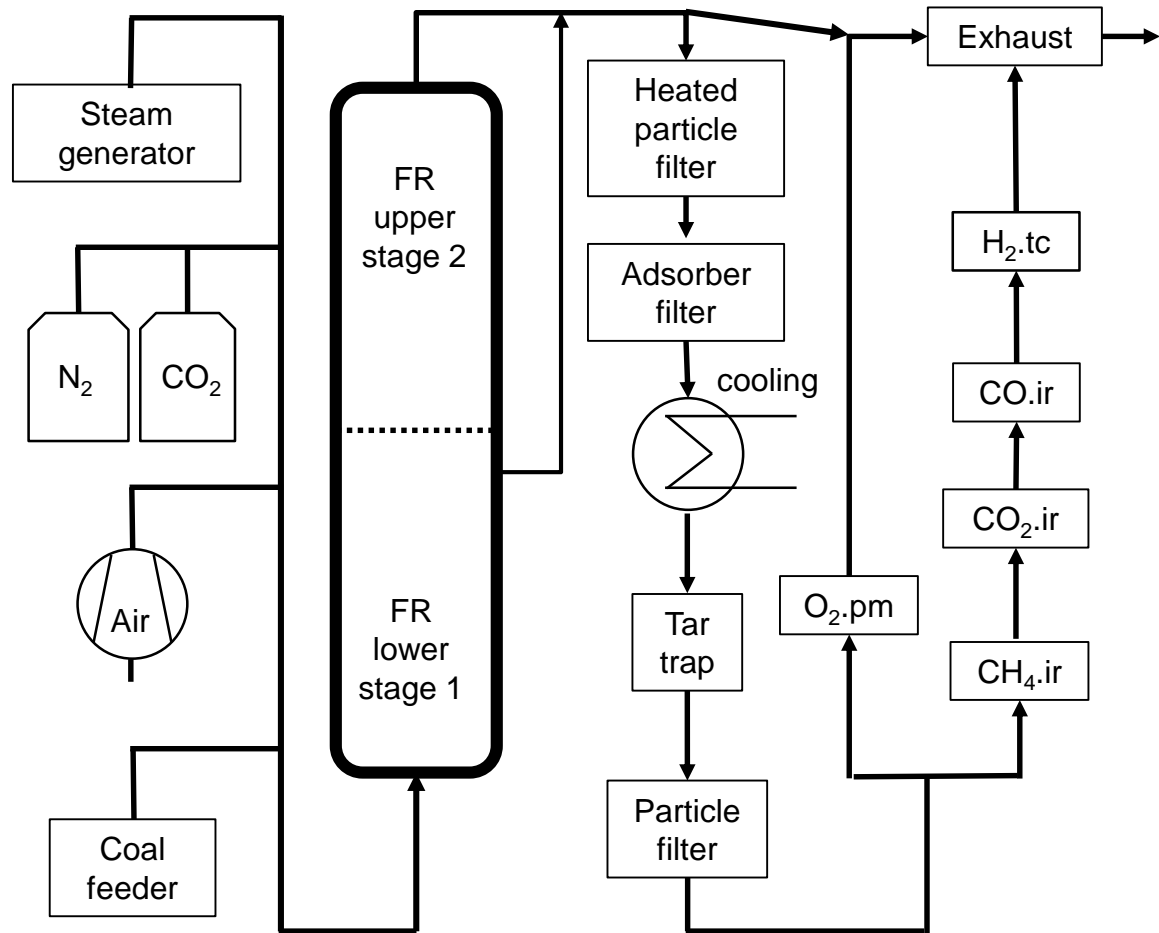
Fuel injection

- Solid fuels: upper or lower stage
- Gaseous fuels: mixed with fluidization gas

25 kW_{th} rated power

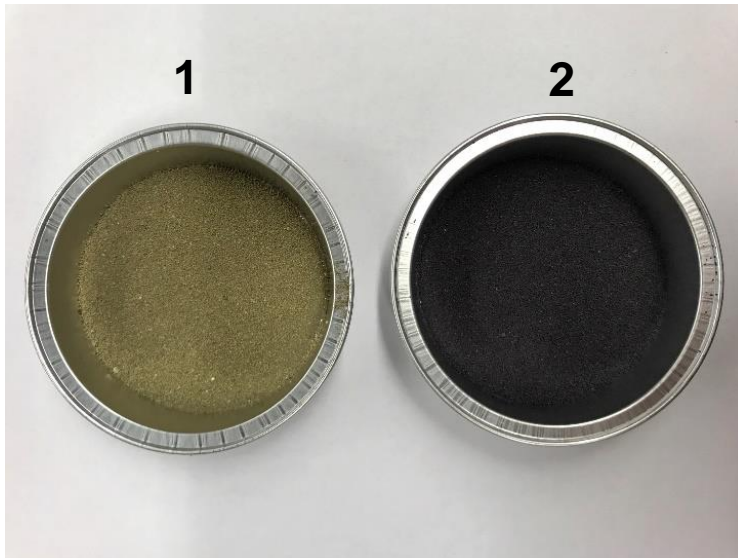
Gas measurements

- CH_4 , CO , O_2 , CO_2 and H_2 detected in FR exhaust
- Possibility to measure after first stage of FR
- CO_2 and O_2 measured at AR exhaust
- CO_2 concentrations include contributions from fluidization and injection gas



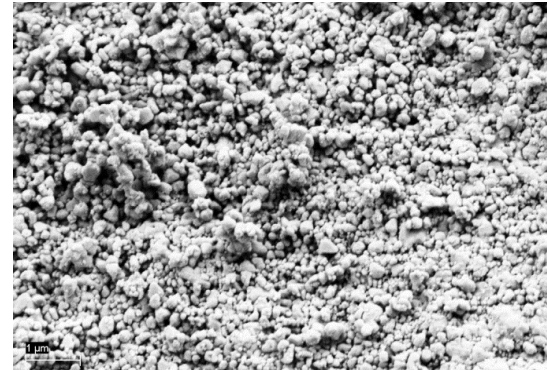
Oxygen Carrier

Molecular composition	$\text{CuO}/\text{Al}_2\text{O}_3$
Density [kg/m^3]	4094
Bulk density [kg/m^3]	1027
Oxygen carrying capacity R_0 [wt. %]	1.50

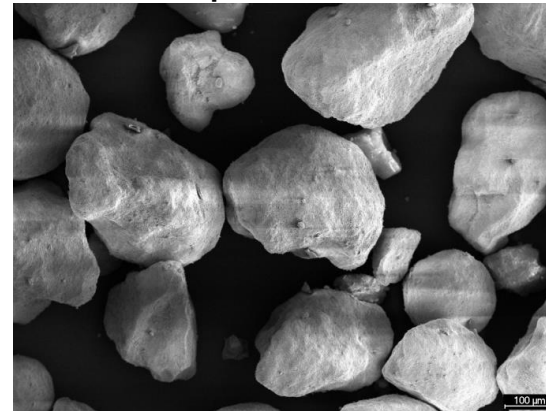


1 = Oxidized carrier
2 = Reduced carrier

↔ 1 μm



↔ 100 μm



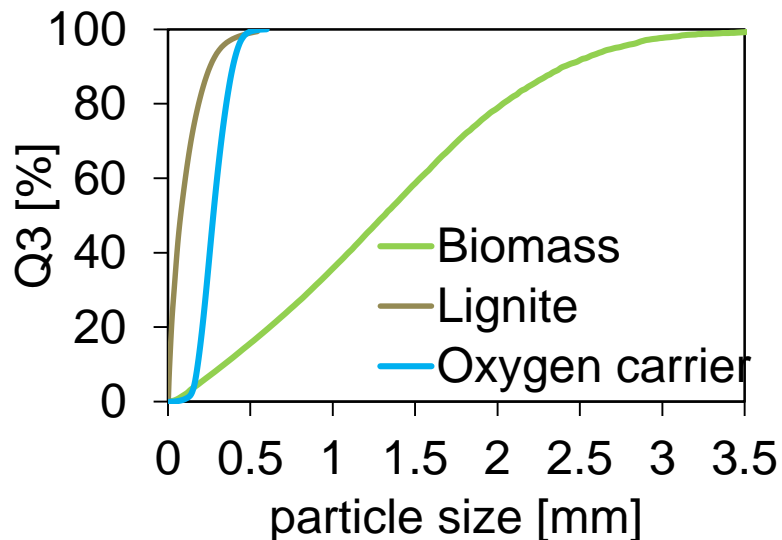
Experimental Pilot Plant

Materials in the pilot plant

German hard wood biomass (shredded pellets)

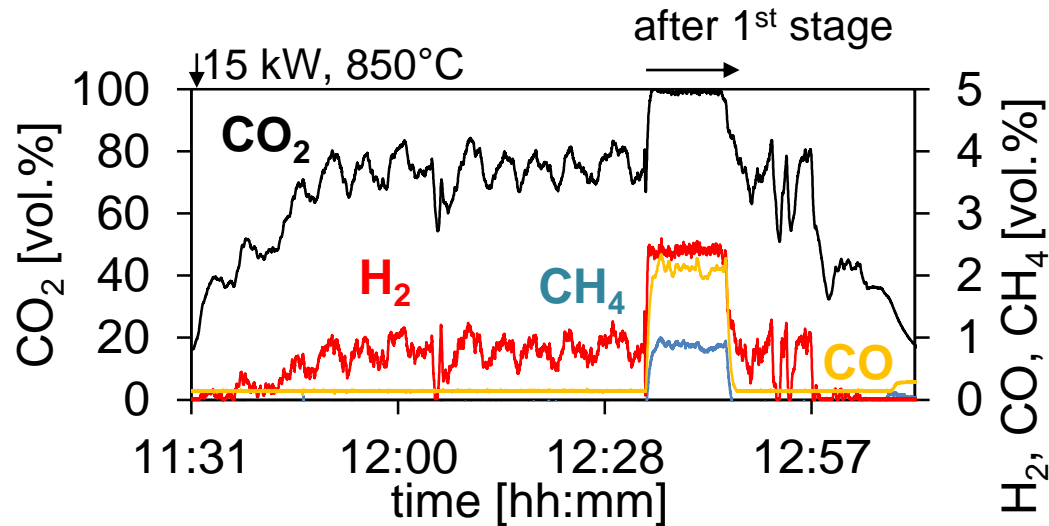


Rhenish lignite dust



	Biomass	Lignite
Bulk density [kg/m ³]	354	1000
LHV [MJ/kg]	18.7	28
Proximate Analysis		
Ash [wt.% raw]	0.2	4
Moisture [wt.% raw]	8.6	11
Volatiles [wt.% raw]	85.5	45
Fixed carbon [wt.% raw]	5.7	40
Ultimate Analysis		
Nitrogen [wt.% raw]	2.9	0.7
Carbon [wt.% raw]	47.1	59.5
Hydrogen [wt.% raw]	10.3	4.3
Sulphur [wt.% raw]	< 0,1	0.35
Oxygen [wt.% raw]	31.9	20.3

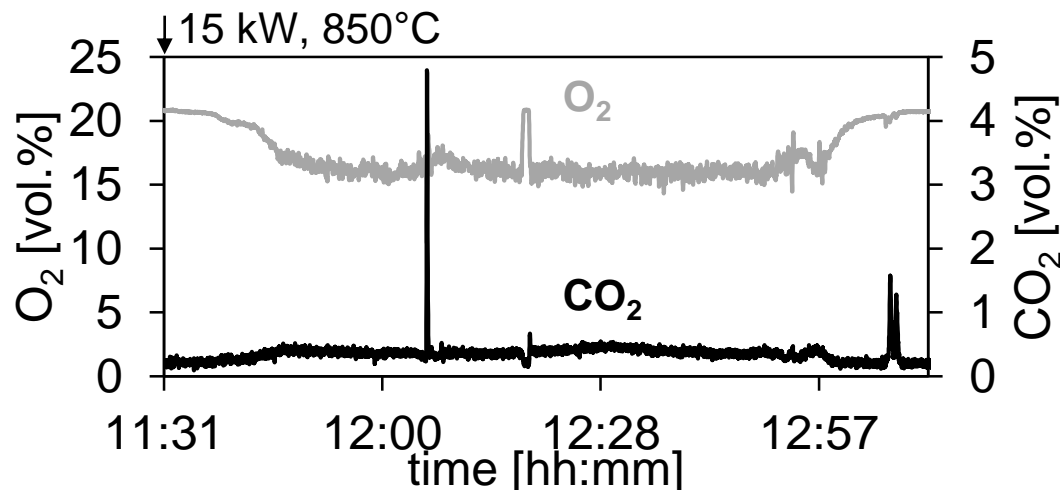
Fuel reactor exhaust gas concentrations (dry)



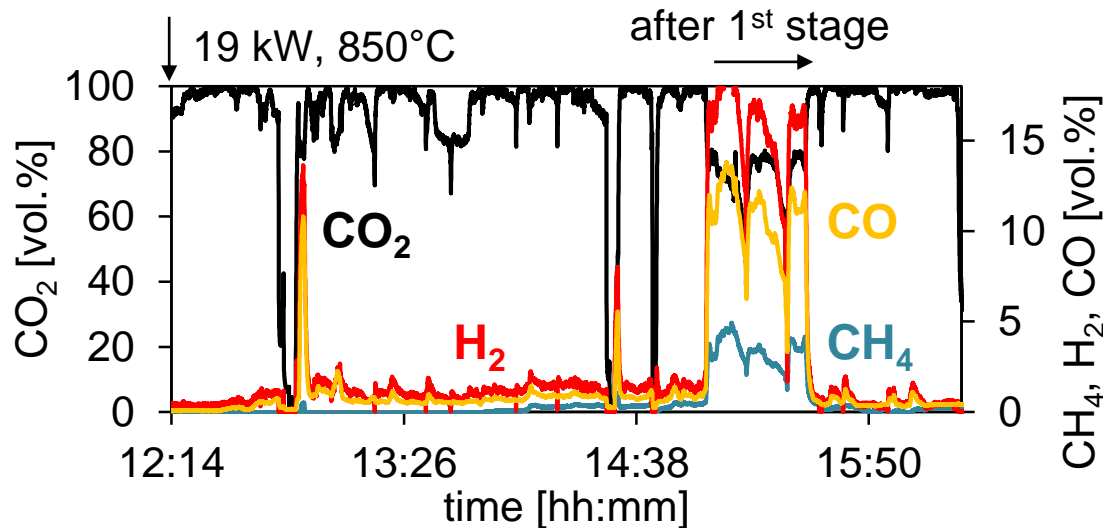
- significant amount of unconverted gases after 1st stage
- low amounts of H₂ after 2nd stage

Overall very good gas conversion, but only after both stages!

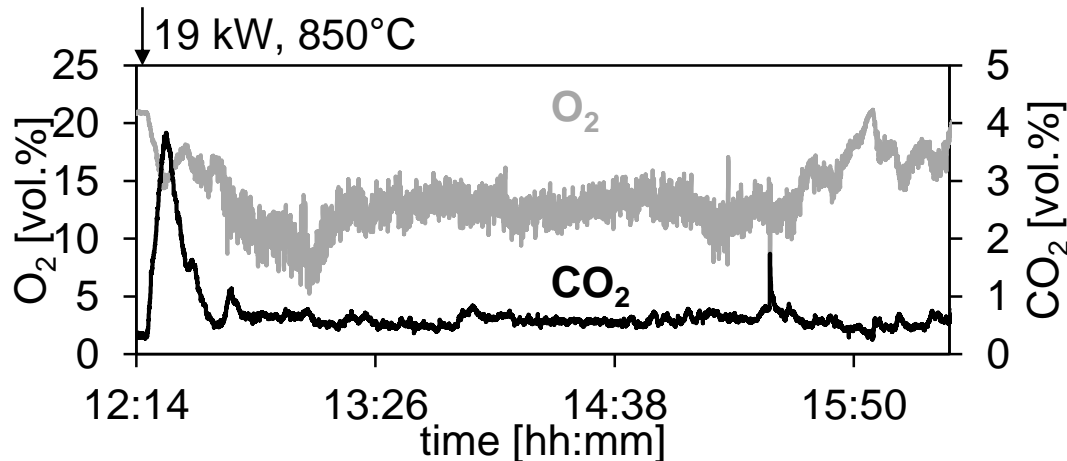
Air reactor exhaust gas concentrations (dry)



Fuel reactor exhaust gas concentrations (dry)



Air reactor exhaust gas concentrations (dry)



- high concentration of unconverted gases after 1st stage
- detectable amounts of H_2 and CO after 2nd stage
- fine particles seem to convert in the upper part of the first stage

Overall very good gas conversion, but only after both stages!

	Fuel input	Temperature	η_{cc}	Ω_{OD}	Ω'_{OD} 1 stage	X_C
Biomass	15 kW _{th}	850° C	0.93	0.016	0.075	0.99
Lignite	19 kW _{th}	850° C	0.97	0.006	>0.25	0.98

- Constantly low oxygen demand (< 3%) for investigated fuels
- Lignite profits strongly from two-stage design (Ω_{OD} 25% → 1%)
- Biomass profits as well from two-stage design (Ω_{OD} 7.5% → 2%)
- Excellent fuel conversion
- Fine particles reduce carbon slip

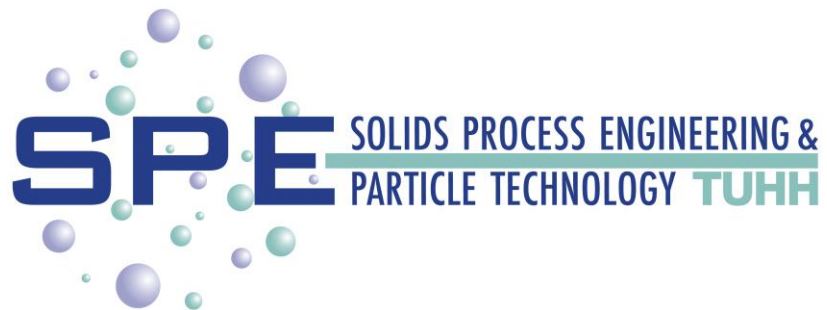
Second stage suitable to reduce oxygen demand during operation with high volatile fuels

- Near 100% solid fuel conversion for both fuels
- After first stage: high shares of unconverted gases were traced
- Combustible gases are almost completely converted after the second stage.
- Reactive char and fine fuel particles reduce carbon slip
- High values of the carbon capture efficiency η_{CC} :
93% for biomass and 99% for lignite
- Lignite and biomass benefit strongly from the two stage design

Two-stage fuel reactor system suitable to tackle one of the main problems of biomass-CLC: the oxygen demand

Current work:

- Reduce pressure drop of the fuel reactor
- Use cheaper oxygen carrier materials



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